

Nanowire Electronics, Photonics and Sensors

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Outline

1. III-nitride nanowires

motivation

growth

device fabrication

electrical/optical properties

role of surface states

3D integration

2. Silicon nanowire transistors/chemical sensors

fabrication

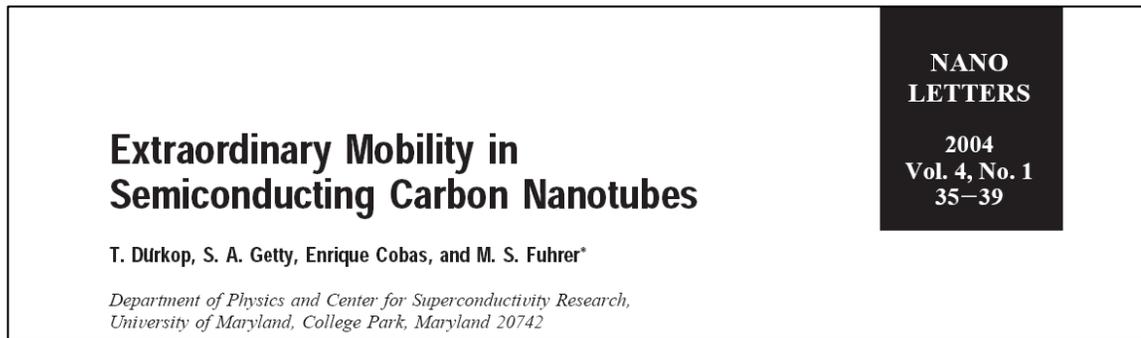
device performance

chemical sensing



Why Nanowires?

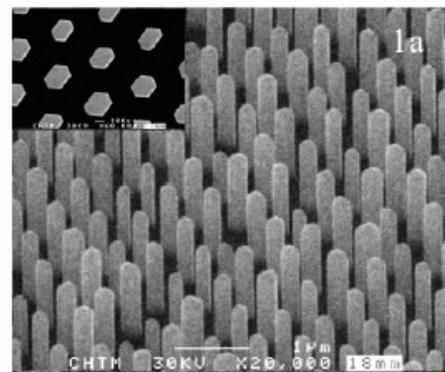
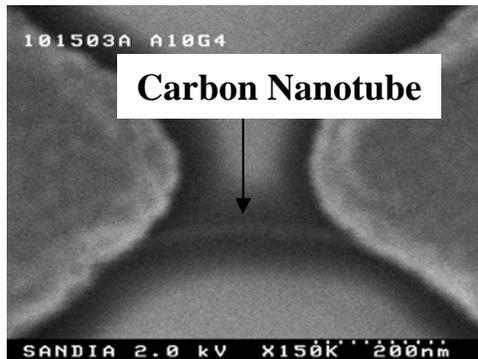
- Unique electrical, optical, thermal properties due to dimensionality (ballistic transport, coulomb blockade, plasmonic resonance, lasing, chemical sensing)



$$\mu_{\text{CNT}} = 79\,000 \text{ cm}^2/\text{Vs}$$

$$(\mu_{\text{Si}} \sim 1500 \text{ cm}^2/\text{Vs})$$

- 40+ materials synthesized as nanowires, lots more possible using similar methods (*Au, Ag, Pd, Sn, Te, ZnO, GaN, Si, Ge, GaAs, InP, SnO₂, TiO₂, C, Ag₂Se, Bi₂Te₃, ...*)
- Ideal building blocks for nanoscale circuits, 3D and Si/III-V integration



A. A. Tai



Why go the nanowire route for III-V semiconductors?

- *Low to zero dislocations* → *speed, lifetime, power*
- *Variety of substrates* → *Si integration, cost*
- *Tens of microns in ~hours* → *3D circuits, lower capac., cost*

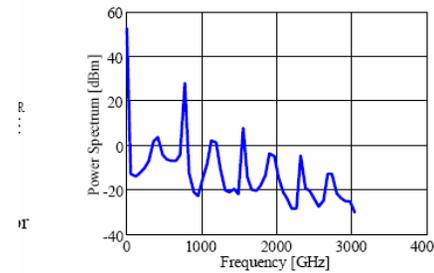
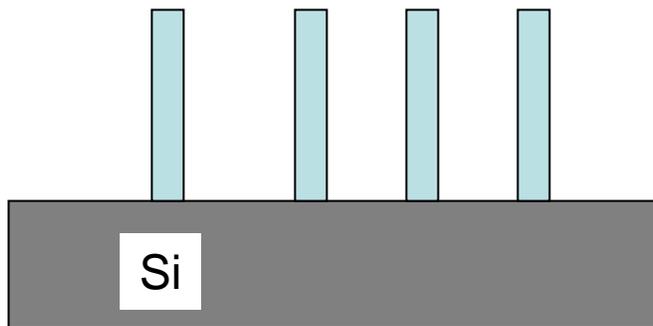
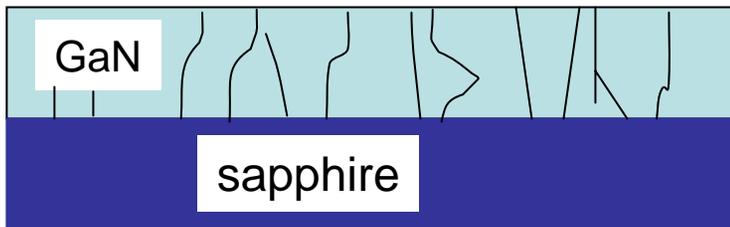


Fig.3. Simulated output power spectrum of 0.3 μm -thick GaN-based NDR diode oscillator designed for operation at THz frequencies.

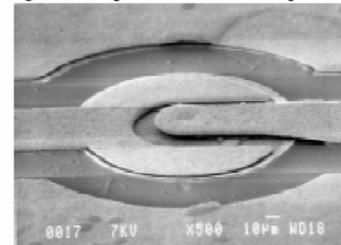
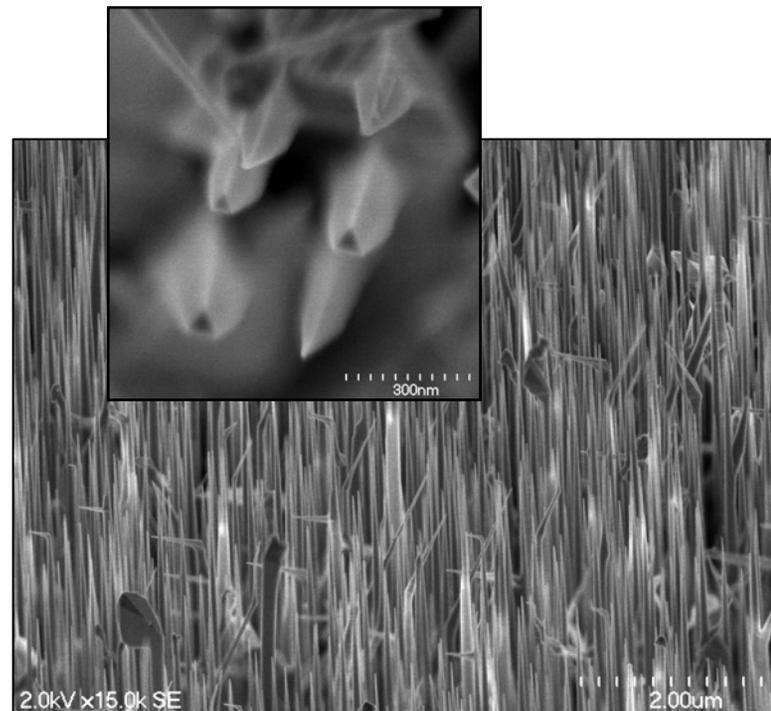
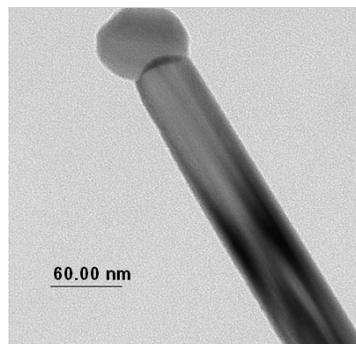
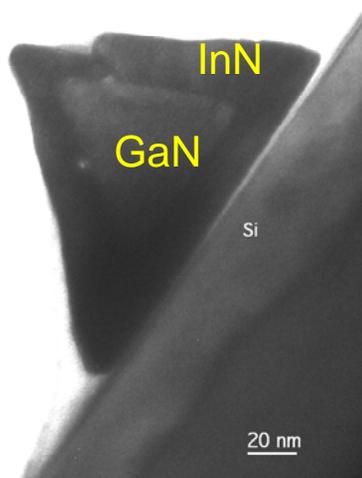
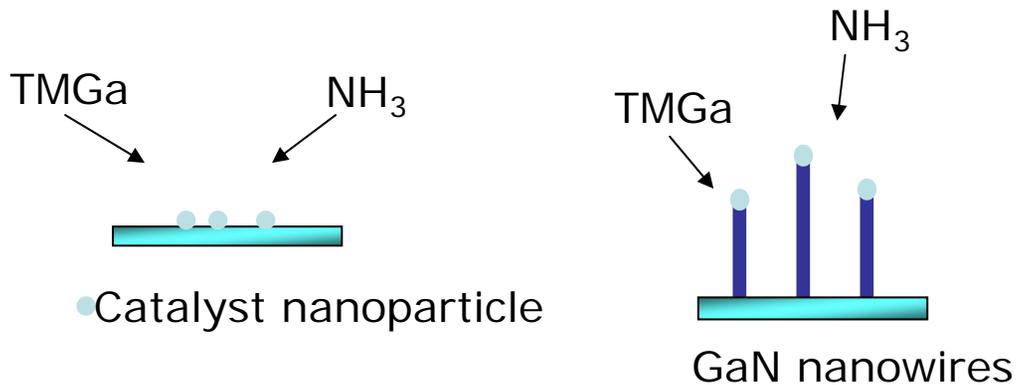


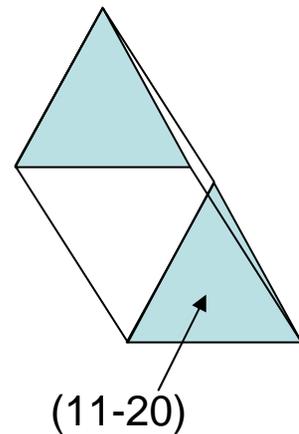
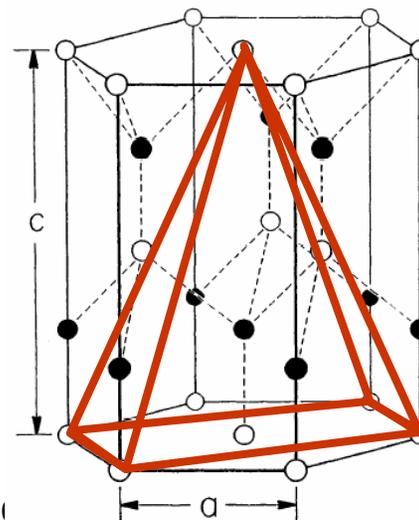
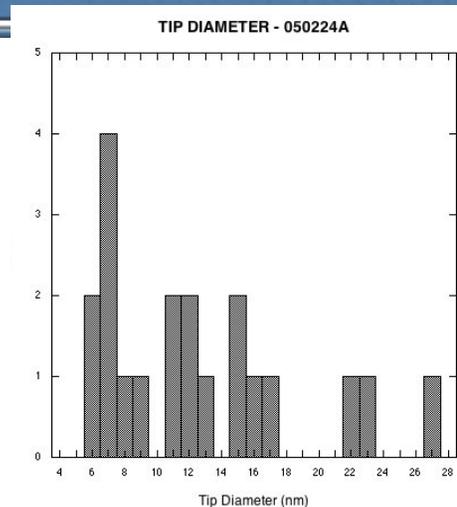
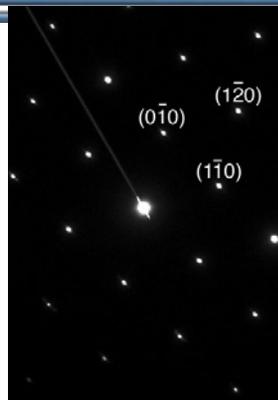
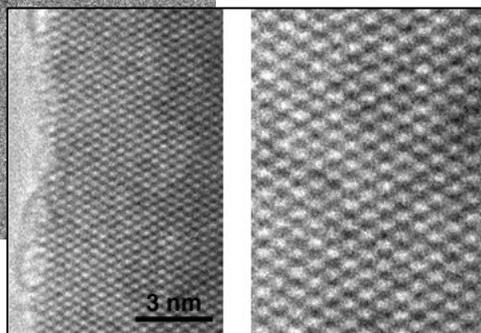
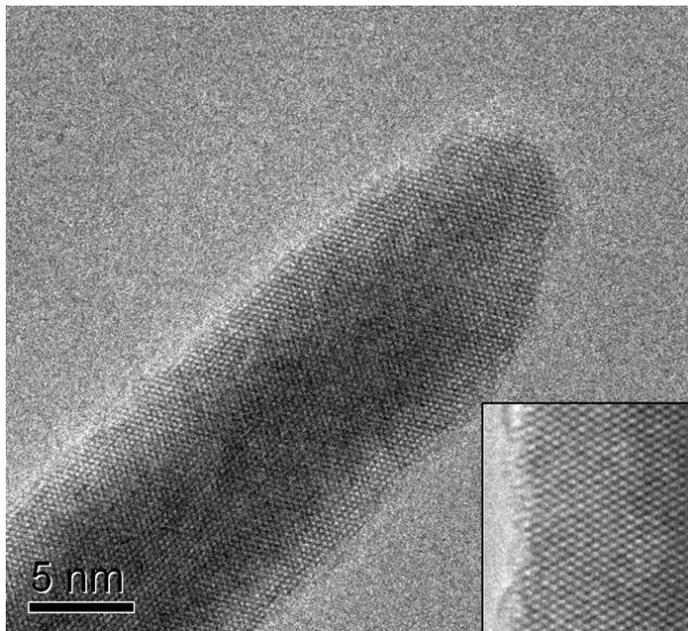
Fig.4. Fabricated GaN NDR Diode with air-bridge interconnects

previous GaN Gunn oscillators failed due to defects, thermal management (Alekseev, Pavlidis, Solid State Electronics 44, 941, 2000)

Metal catalyzed CVD growth of GaN nanowires



TEM of Nanowires 800°C, sapphire r-face



(10-11)

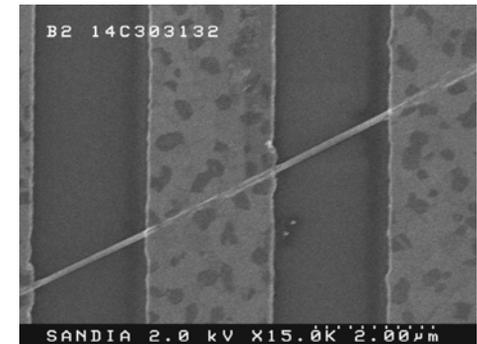
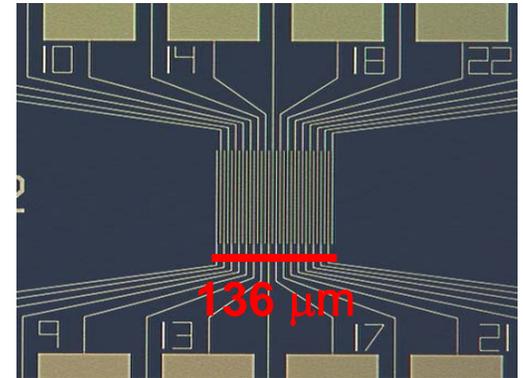
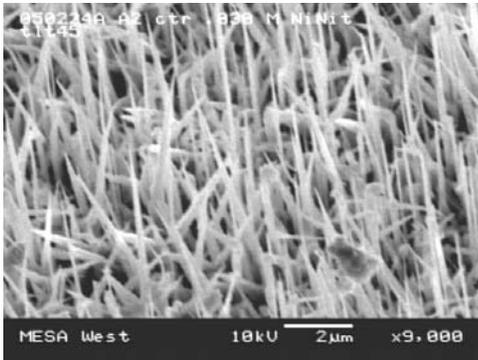
- Aligned nanowires single crystalline and
- Primary growth direction $[010]$

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Device fabrication

- Sonication to transfer nanowires from growth substrate to test substrate
- UV aligner and lift-off to pattern individually addressable electrodes

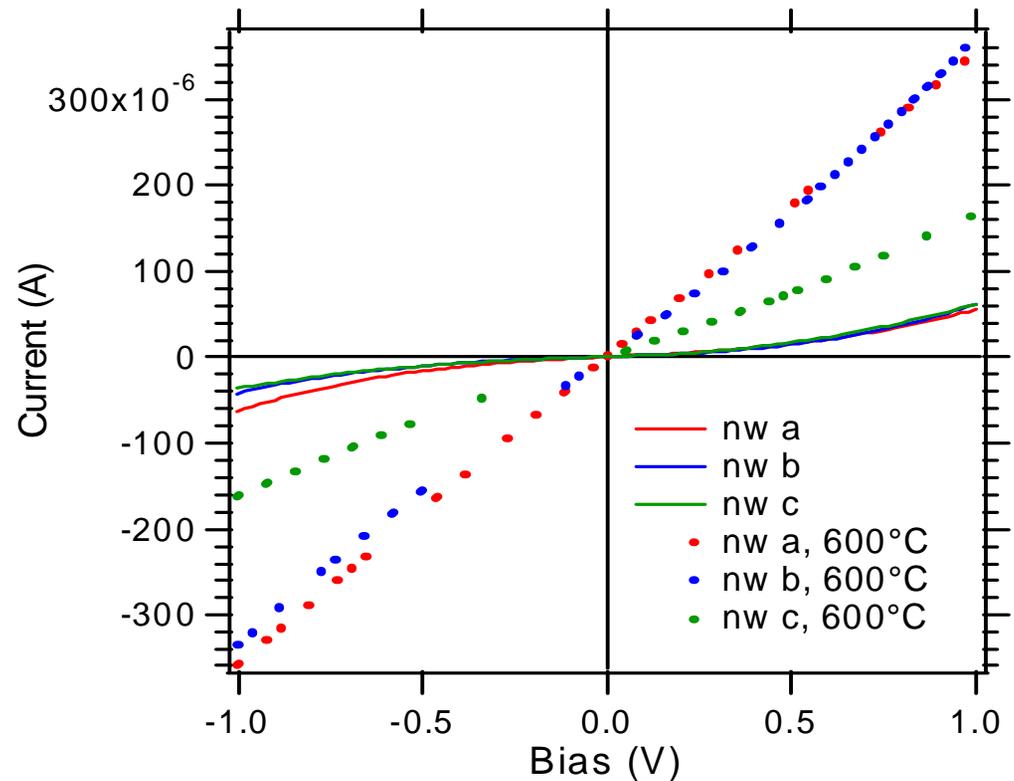
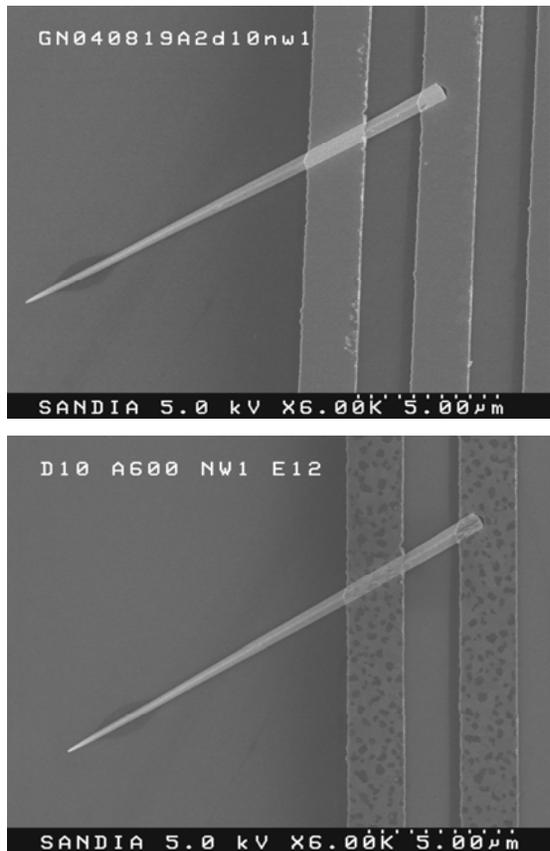


- Simple method of device fabrication resulting in high yield of testable nanowires
- Alternative to E-beam writing

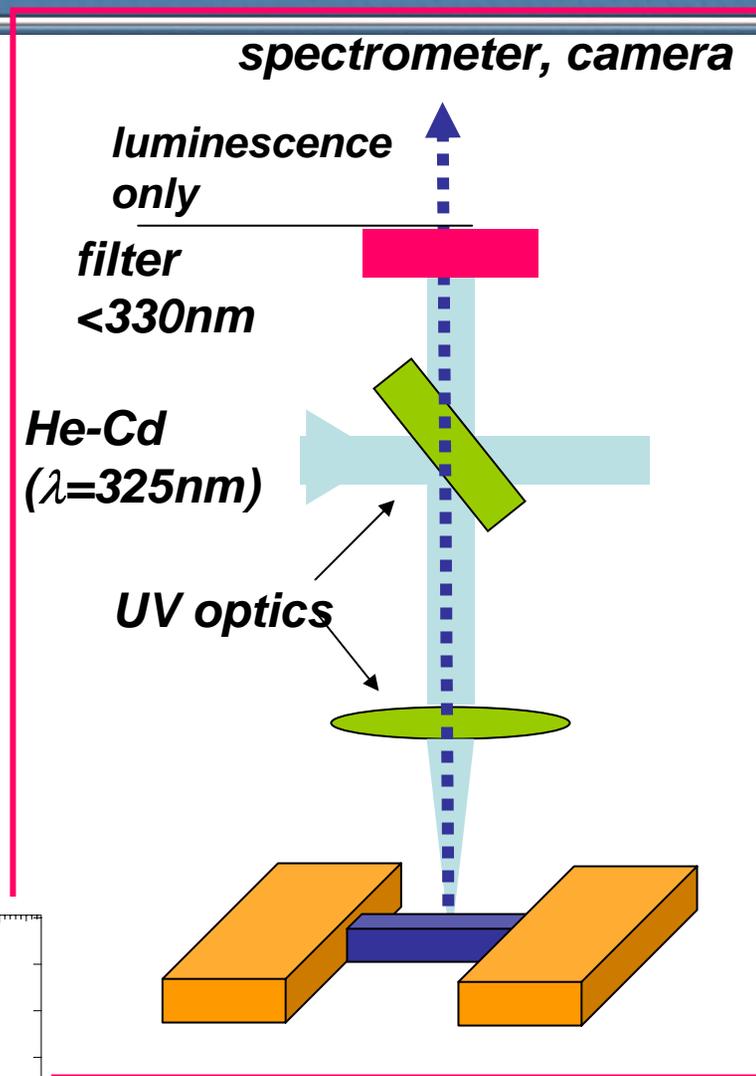
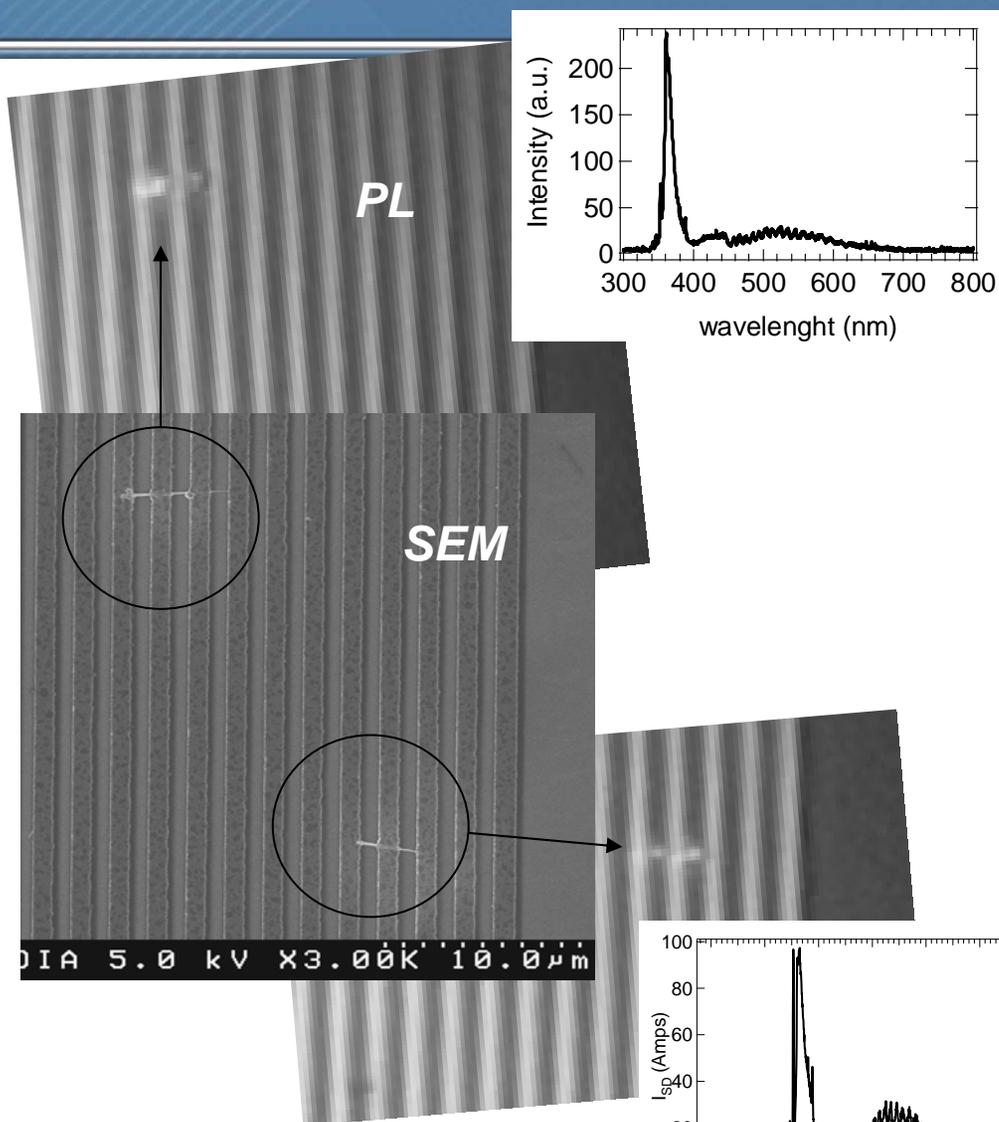


Ohmic contact formation

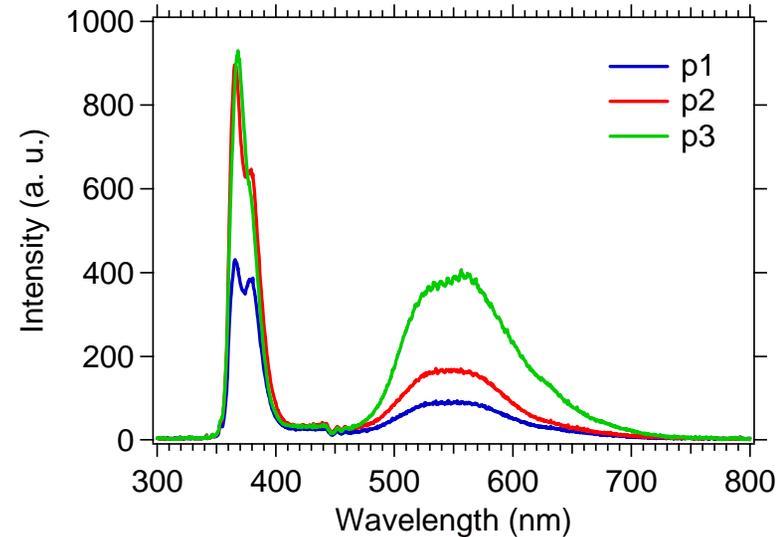
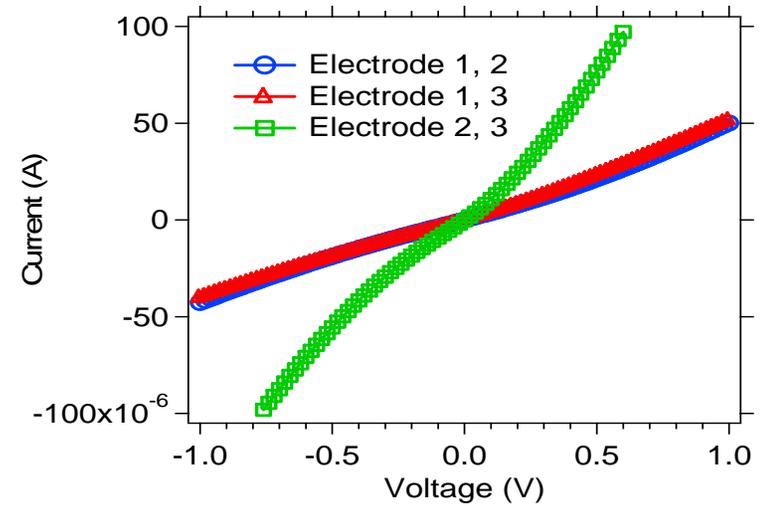
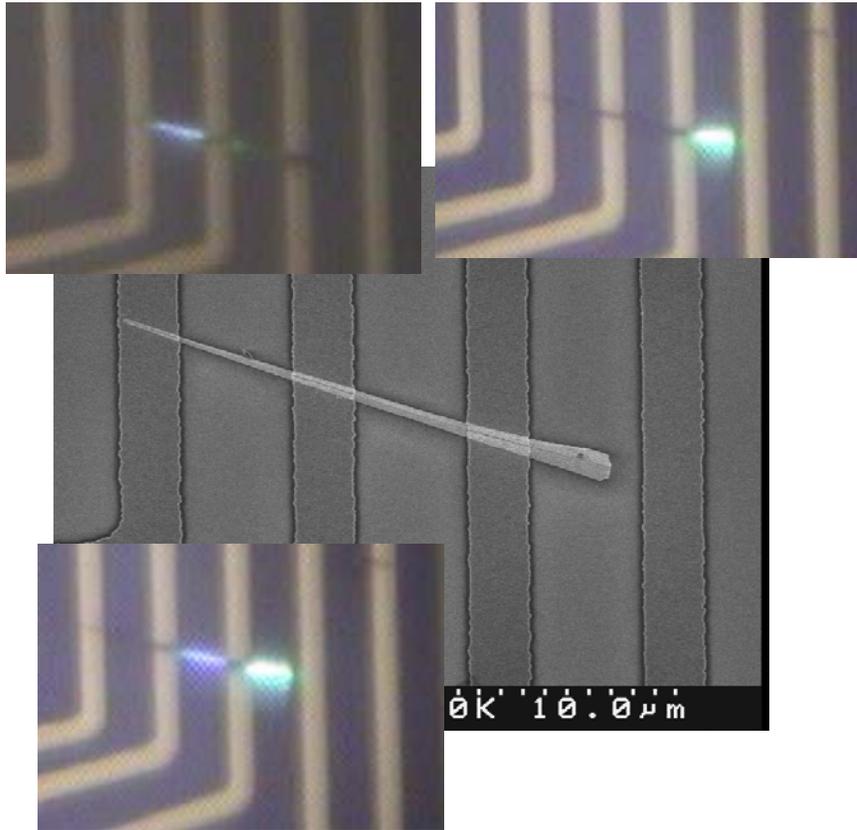
- Ni/Au contacts mostly resulted in Schottky contacts
- After 20 minute 600°C anneal, all contacts become less resistive & most contacts become Ohmic



Single Nanowire PL Imaging/Spectroscopy System

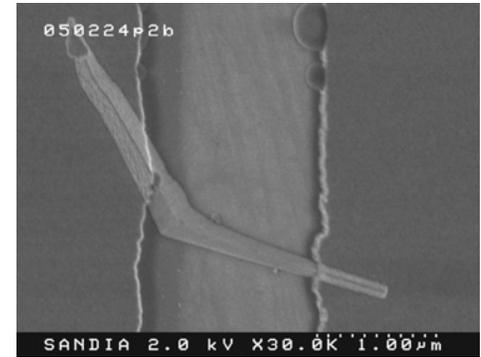
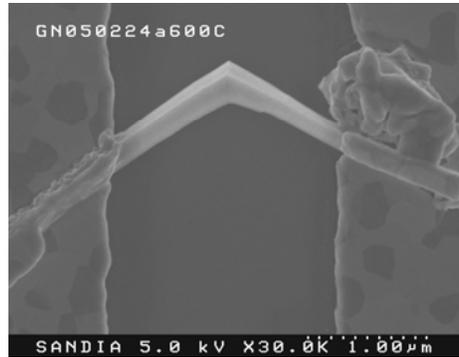
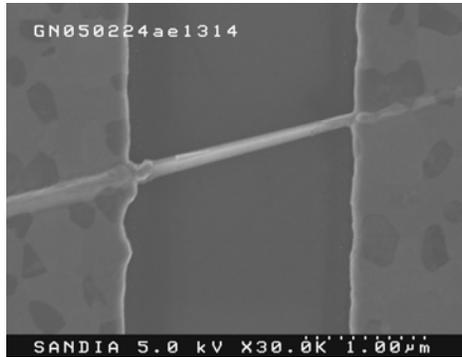


Spatially resolved PL imaging, spectra, I/V

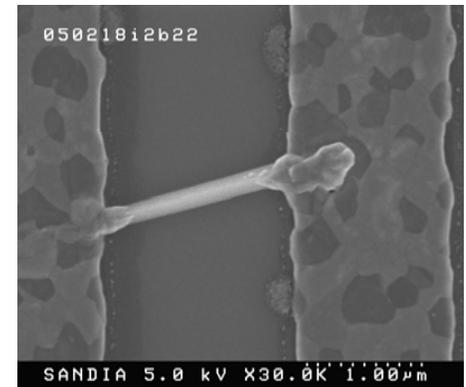
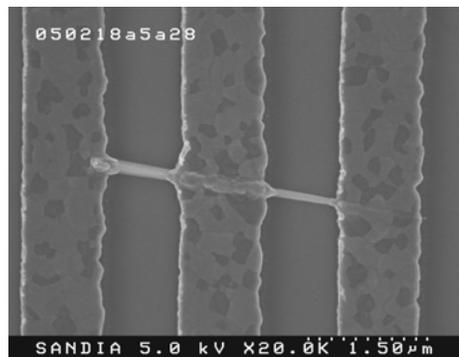
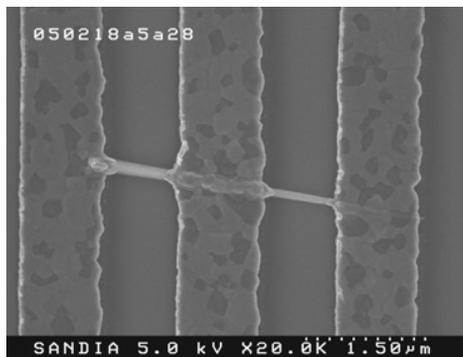


Morphology of 'integrated' nanowires: planar defect obvious in 800°C nanowires

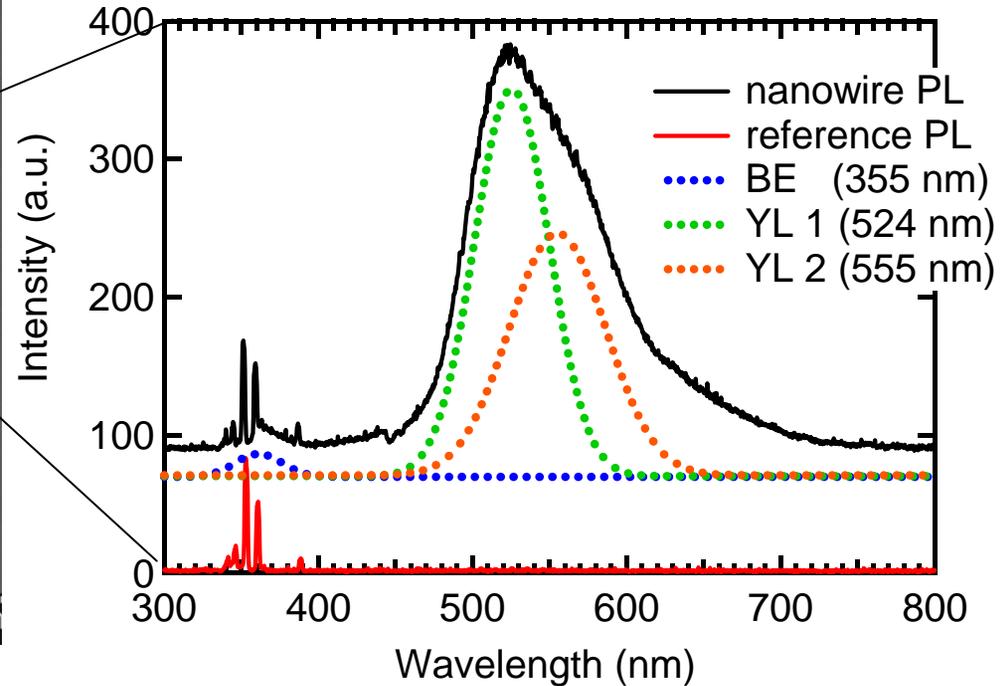
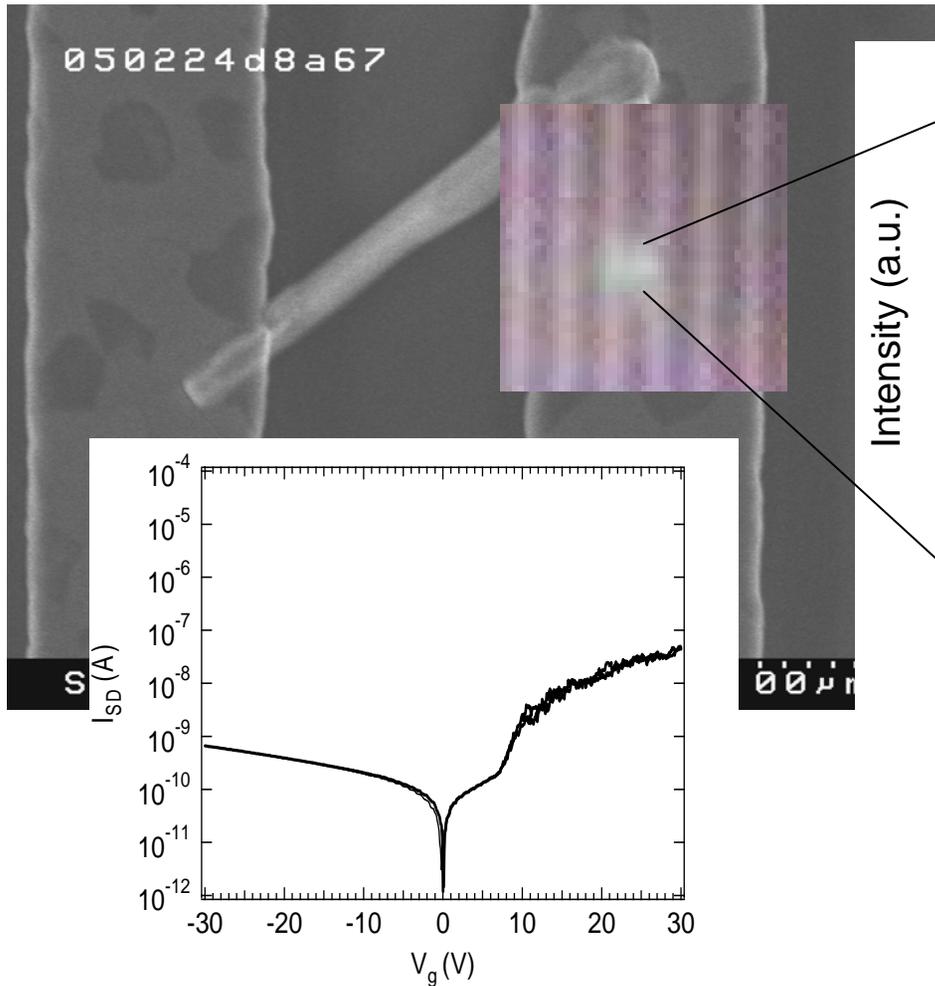
800°C



900°C



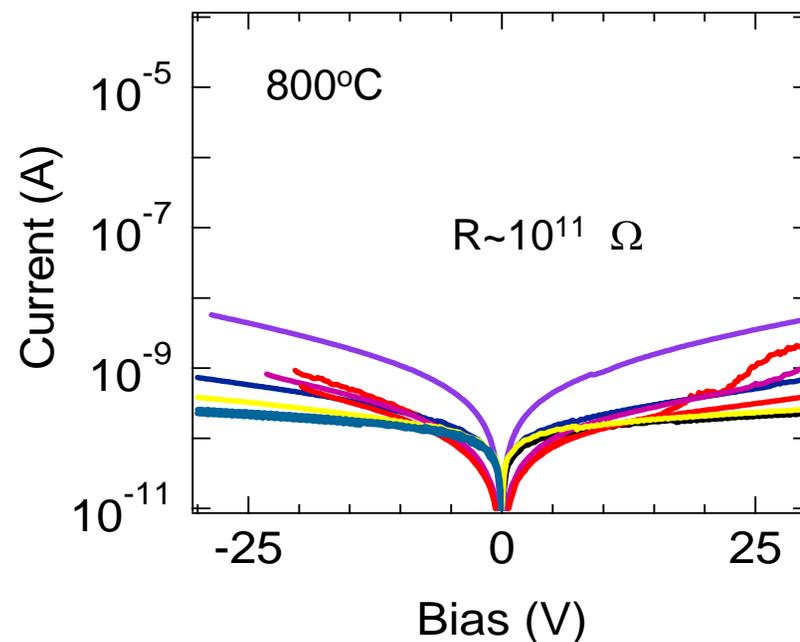
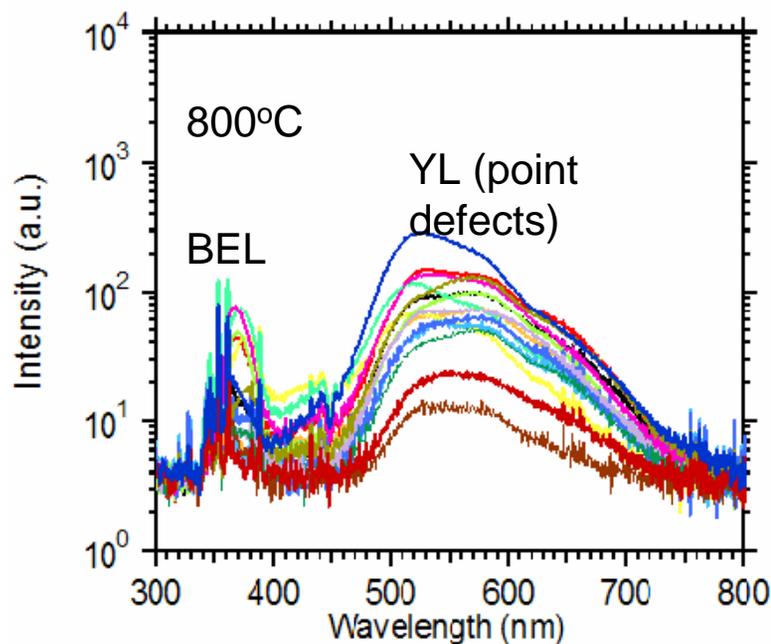
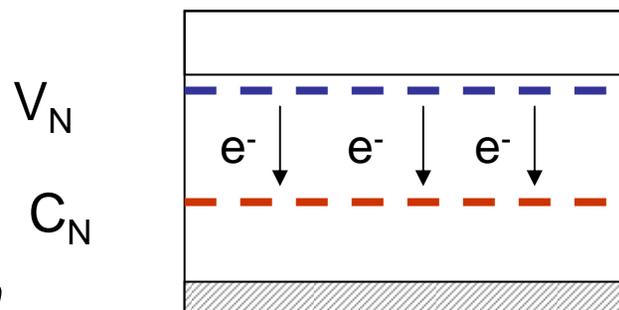
Nanowires grown at 800°C: PL intensity weak, at least 2 peaks in YL, high R



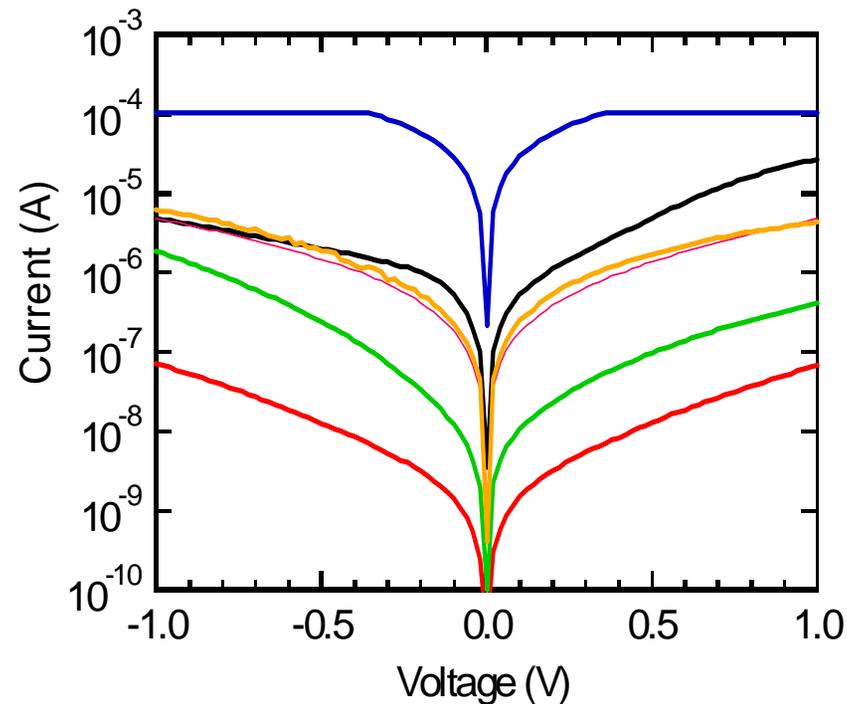
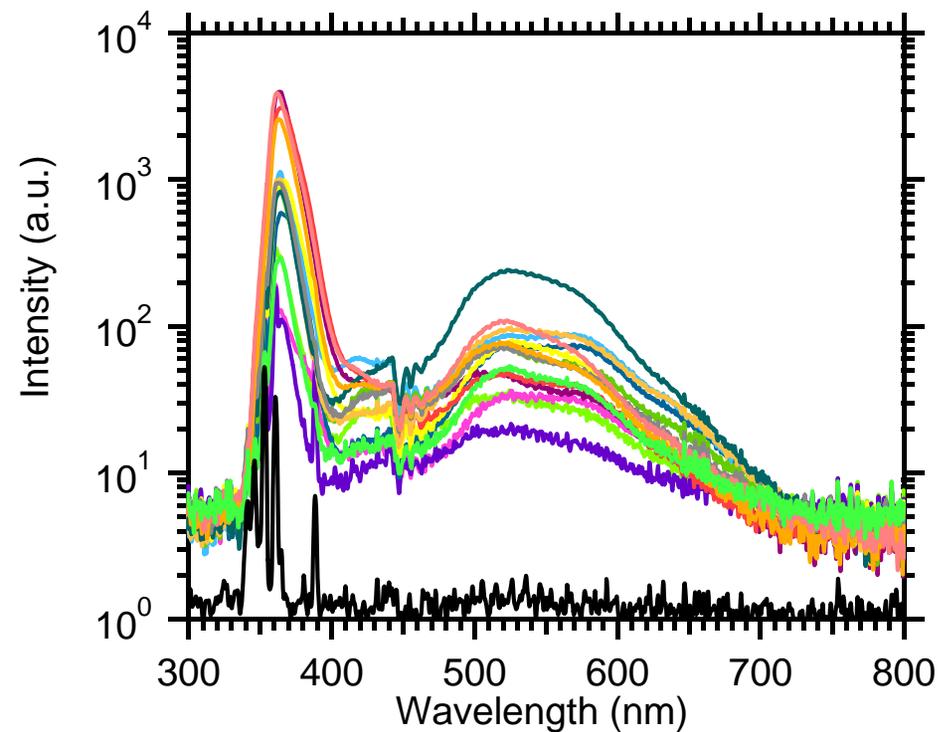
Effect of growth T on PL, I/V: GaN at 800°C

- PL dominated by YL
- Resistance $\sim 10^9 \Omega$

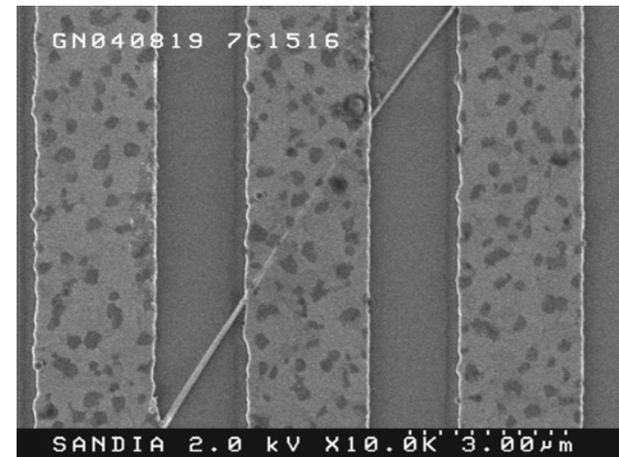
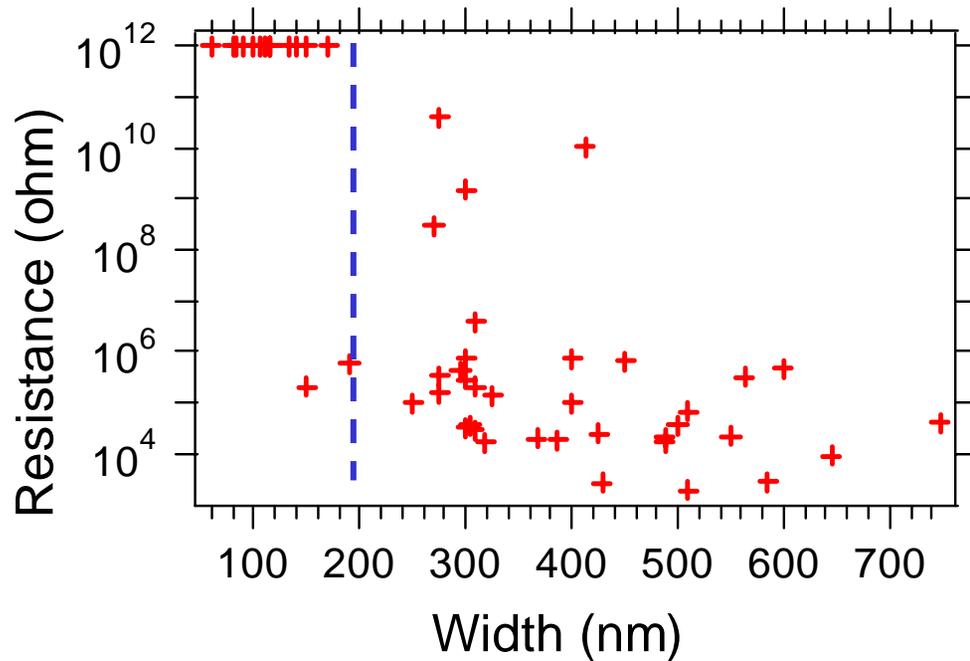
Carbon from TMGa may be deep level acceptor; lead to carrier compensation



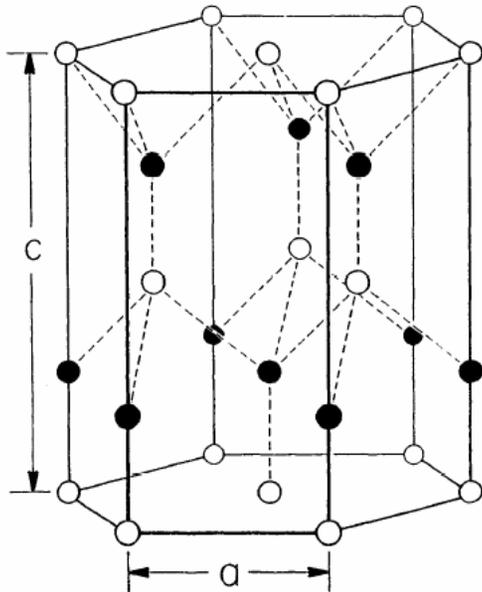
Effect of growth T on PL, I/V: 900°C-> BEL dominates, $R \sim 10^5 \Omega$



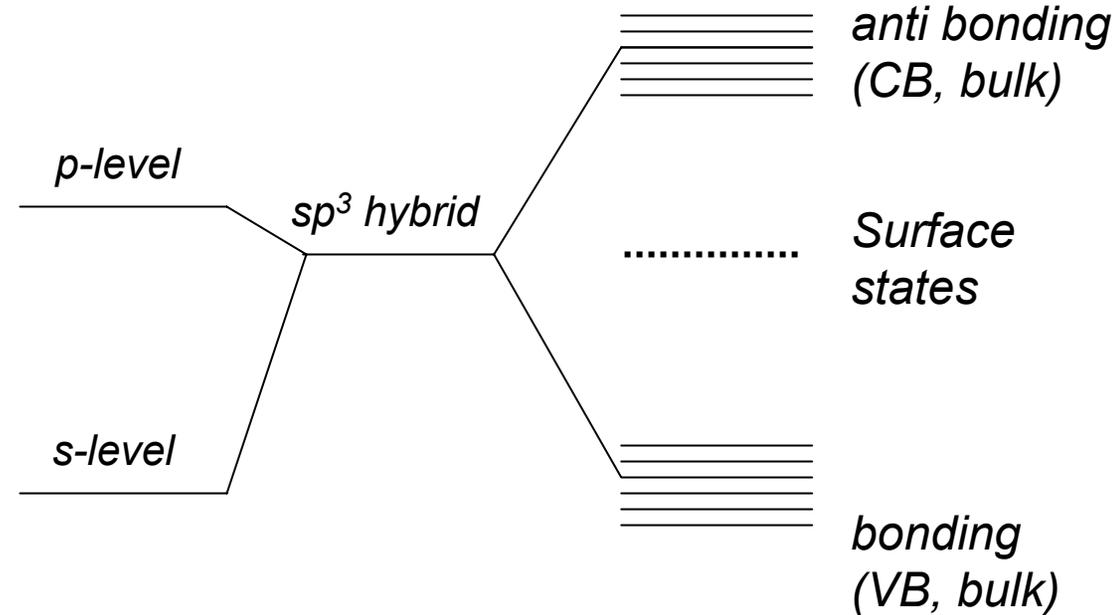
Bellow ~200 nm, GaN nanowires grown at 900°C are insulating!



A short detour....creation of a surface creates broken, 'dangling' bonds



Periodic arrangement of atoms in solid leads to allowed energy bands, gaps

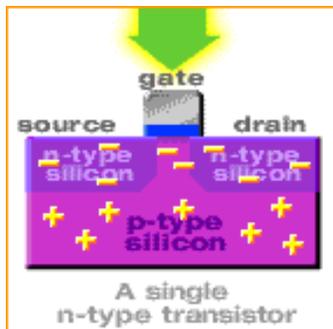
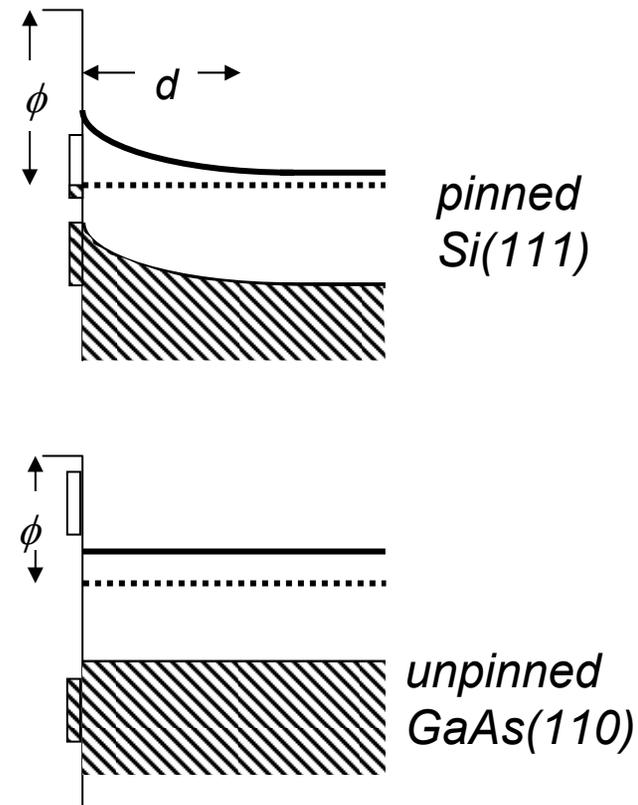
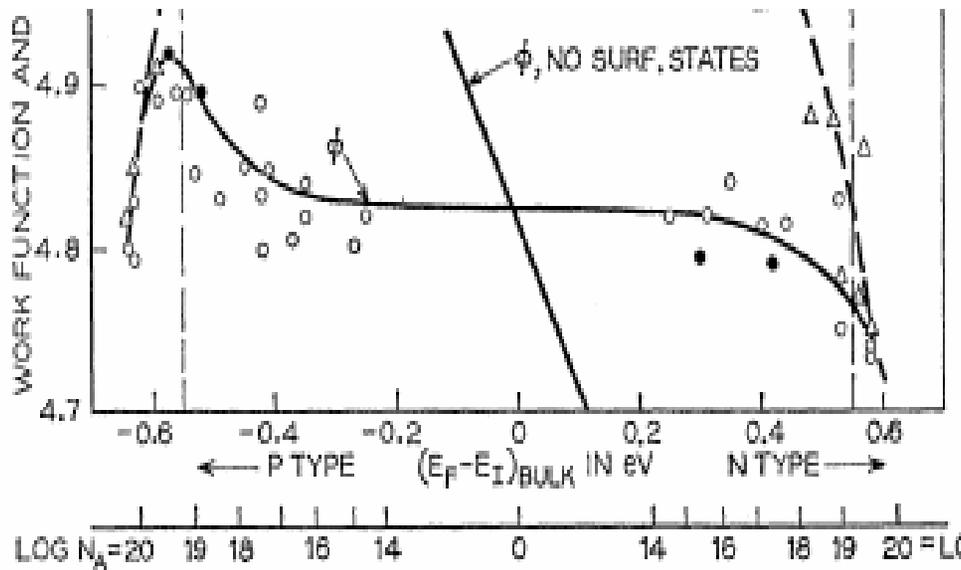


- Creation of a surface breaks the order, leads to energy states in the forbidden gap
- Surface reconstruction, adsorbates determine position and density of gap states



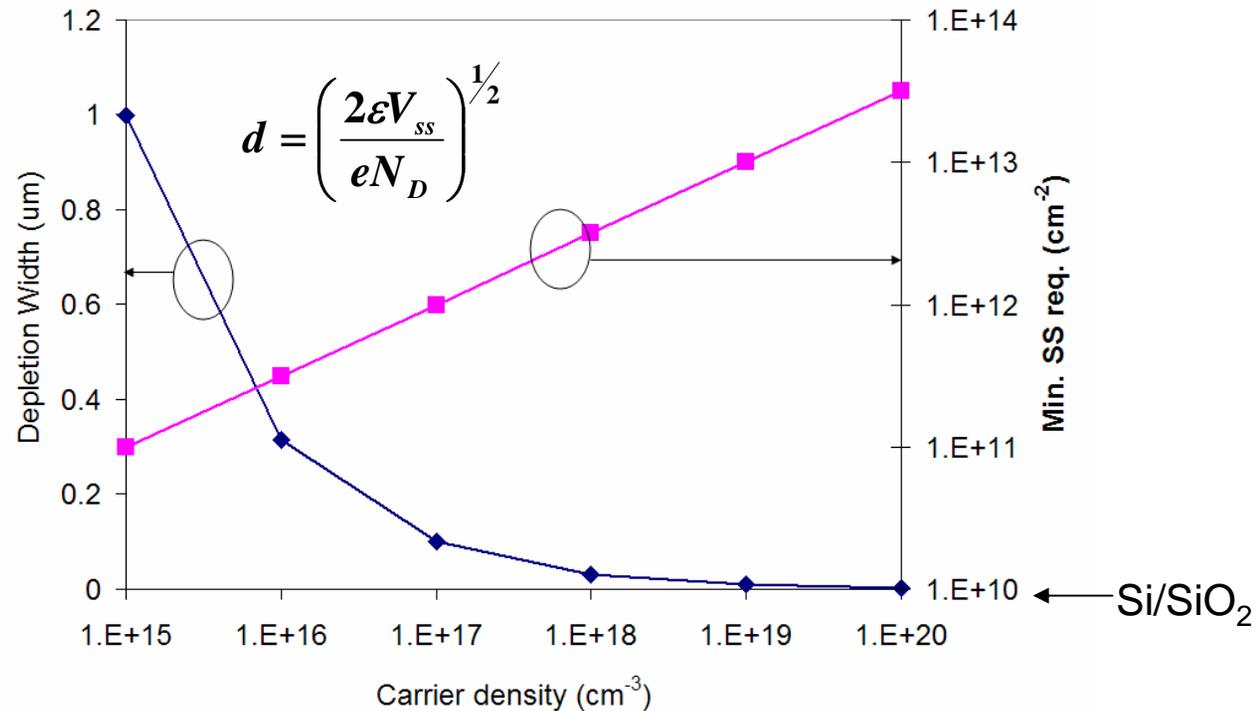
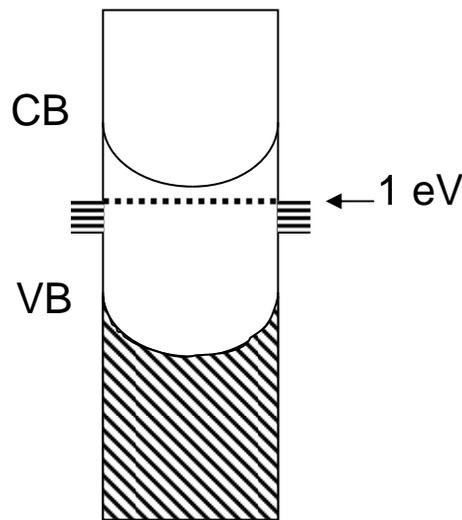
Most semiconductor surfaces have 'pinned' Fermi level due to surface states

Work function of cleaved Si(111) is only weakly dependent on doping

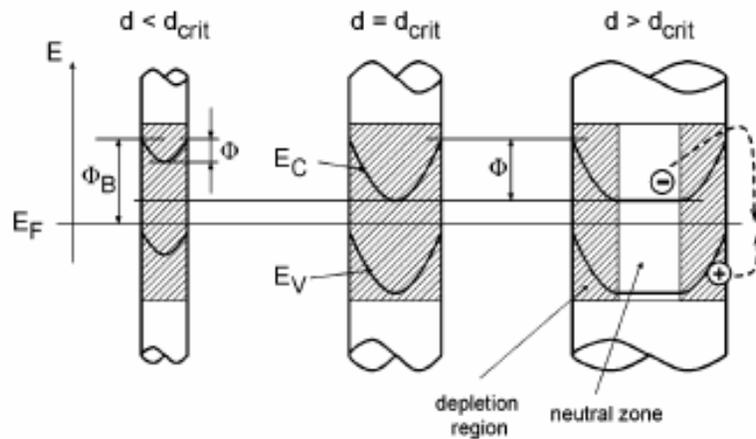


Fermi level pinning not an issue for bulk Si devices, since SiO_2 passivates dangling bonds

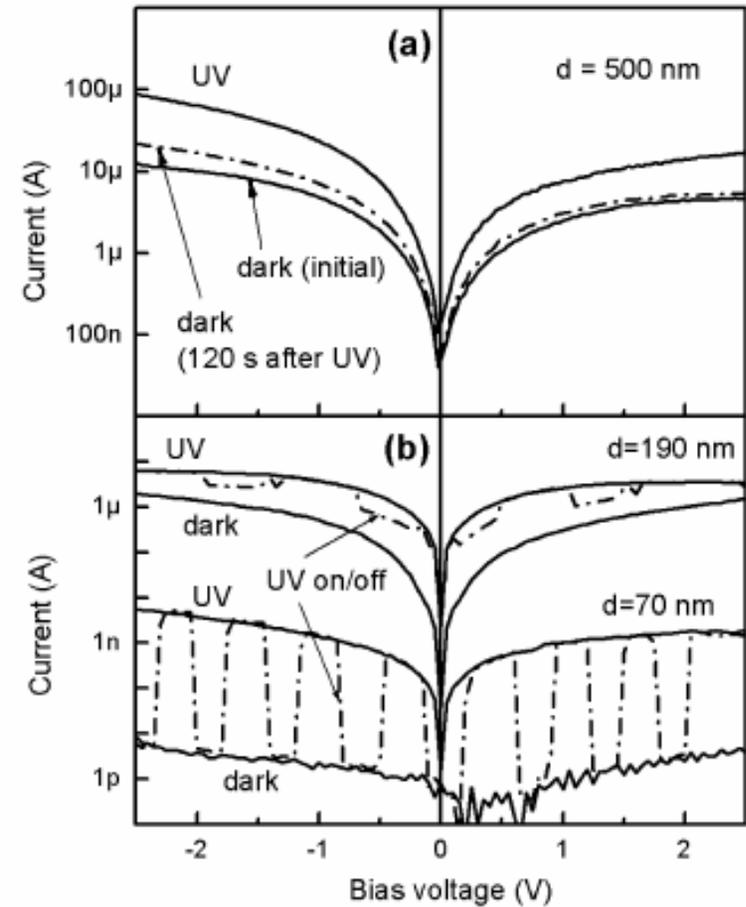
At $[e^-] \sim 10^{17}/\text{cm}^3$, $E_{ss} \sim 1\text{V}$, and SS density $> 10^{12}/\text{cm}^2$, complete depletion at diam. of 200 nm



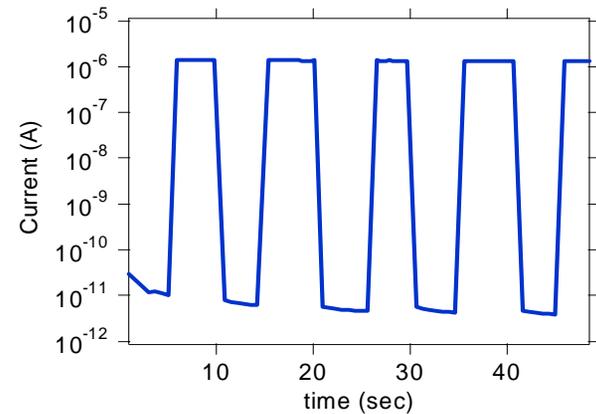
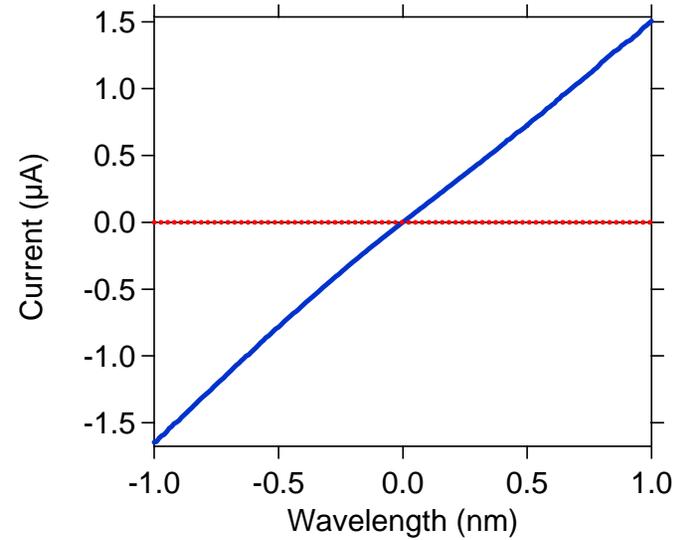
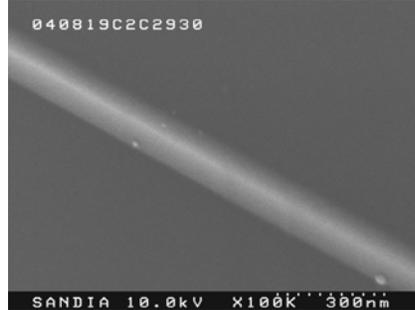
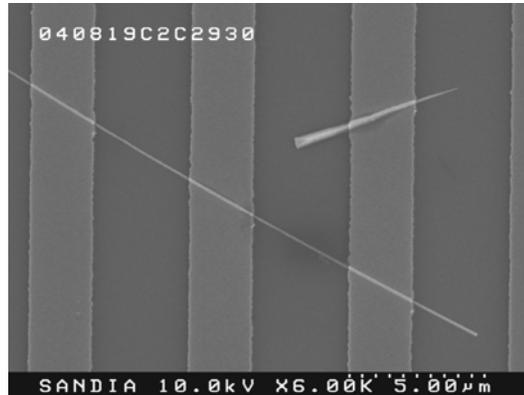
We are not the first to not role of surface depletion on optical, electrical properties of nanowires...



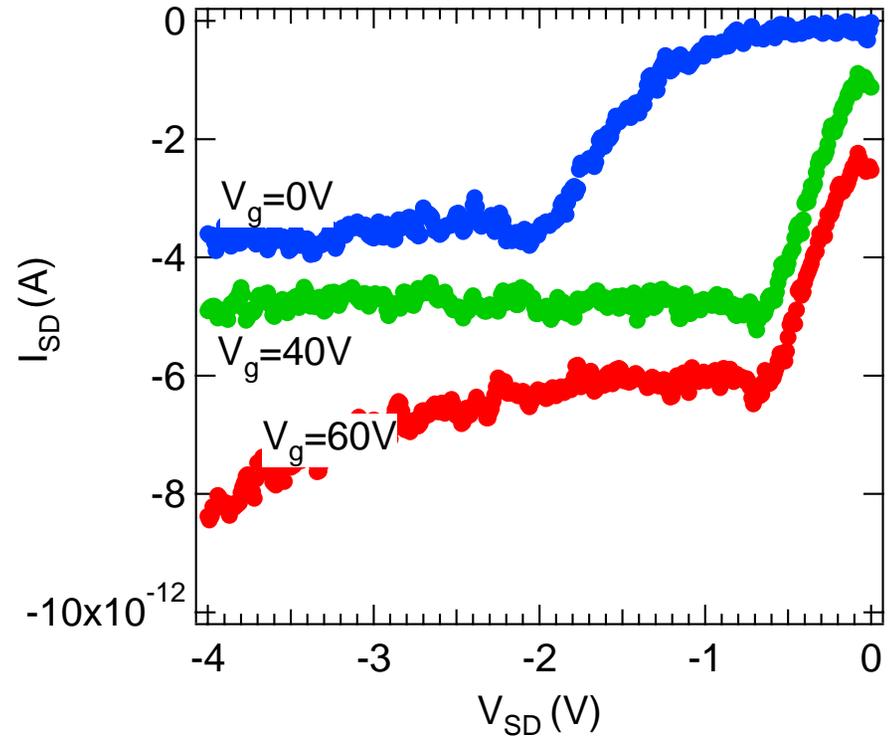
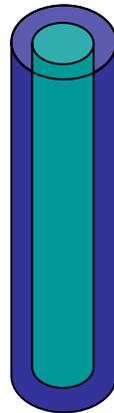
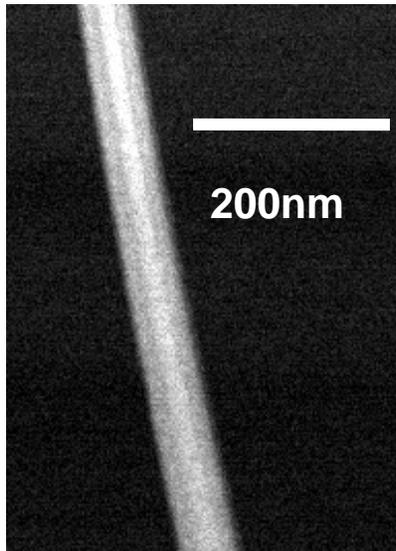
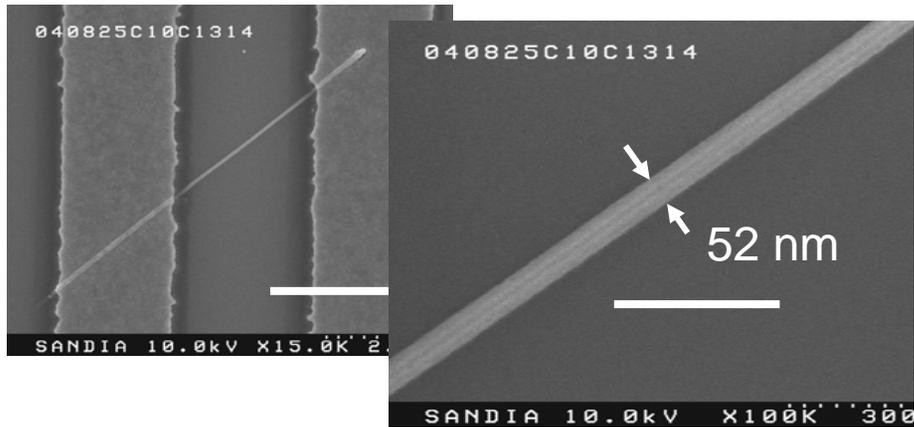
R. Calarco et. al Nano Lett. 5, 2005



The depleted GaN nanowires are promising UV photodetectors, with low dark current, high gain



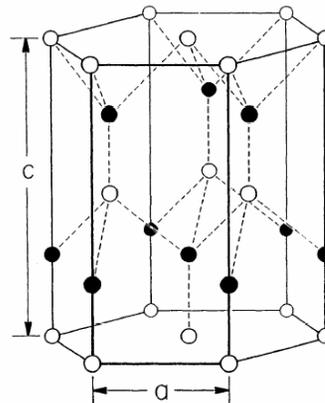
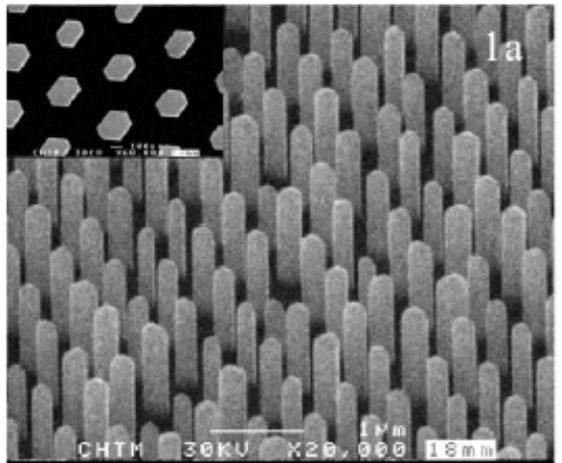
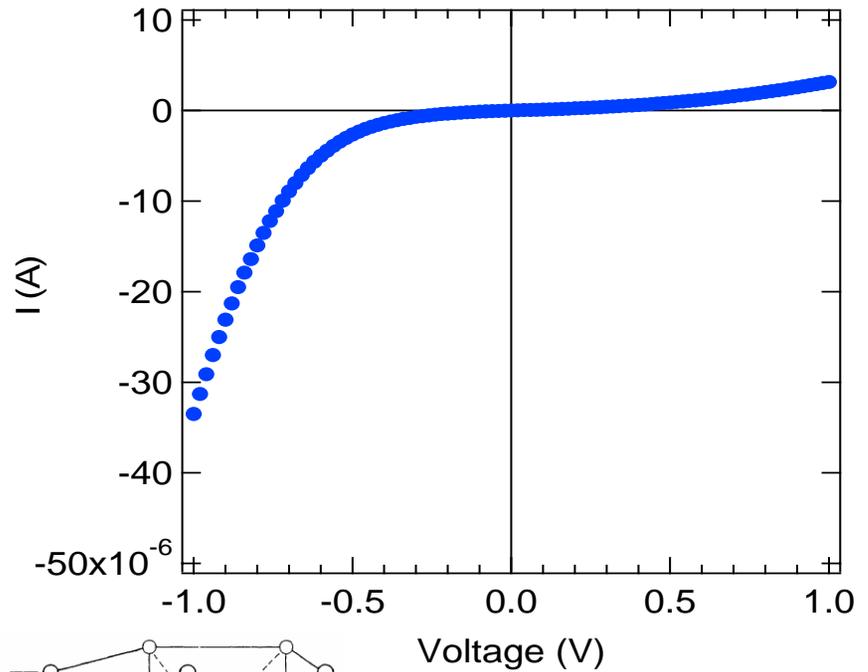
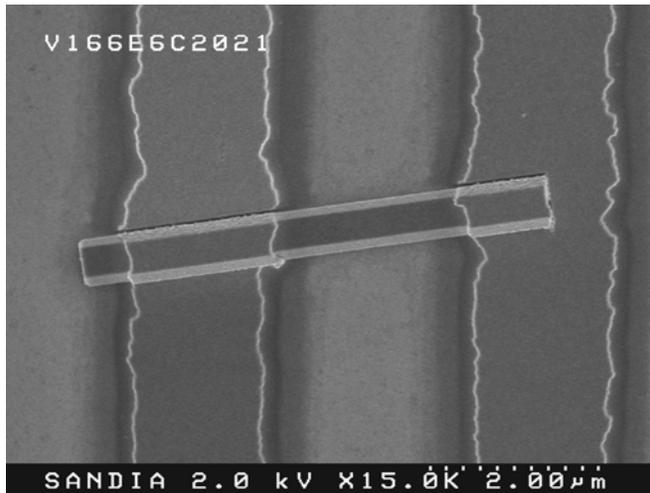
AlN cladding is a possible surface passivation layer for GaN nanowire devices



AlN/GaN core-shell nanowires:
conduction observed even at 50 nm
diam. for some, other are insulating



Another possible approach is to change the surface/orientation of the nanowire (w/S. Hersee, UNM)

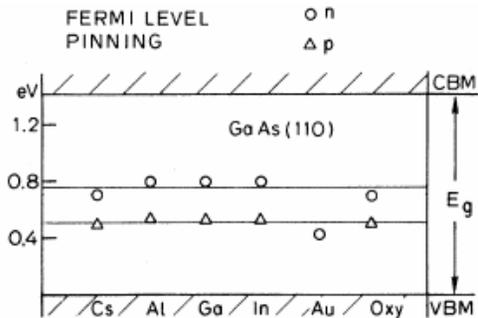


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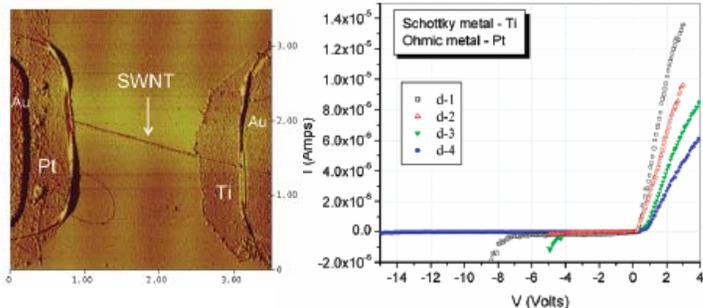


Metal contacts to CNTs and nanowires: Schottky barrier, Fermi level pinning depend on diameter

contacts to bulk semiconductors

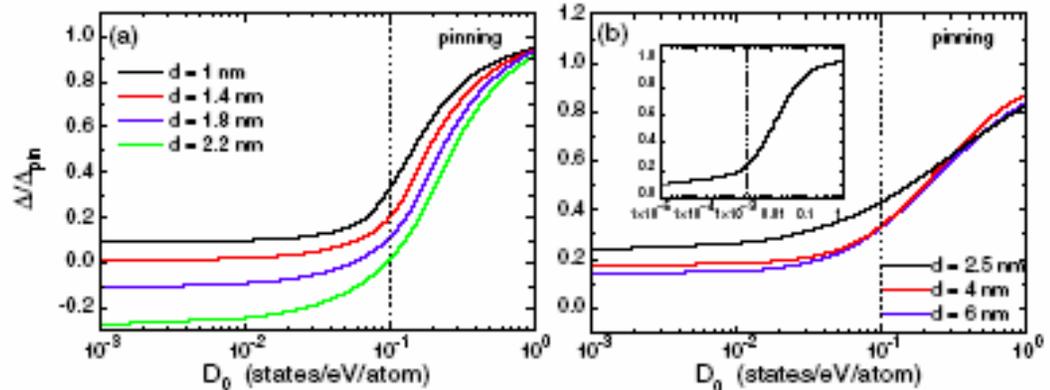
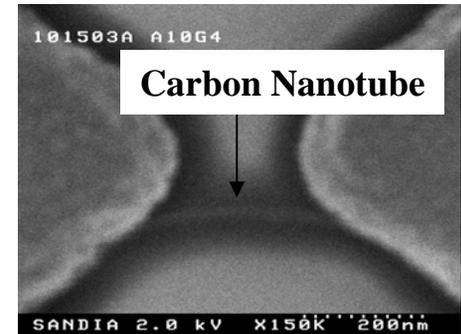
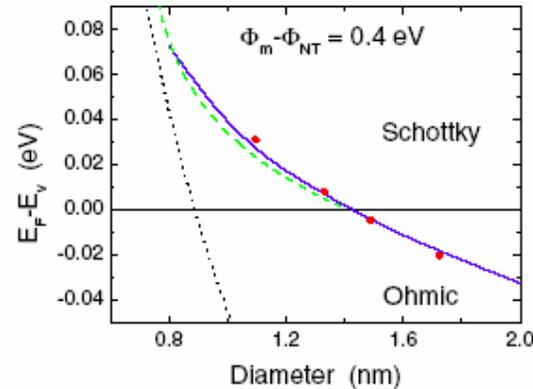


Spicer et al., PRL 44, 1980



P. Siegel, Nano Lett. 2005)

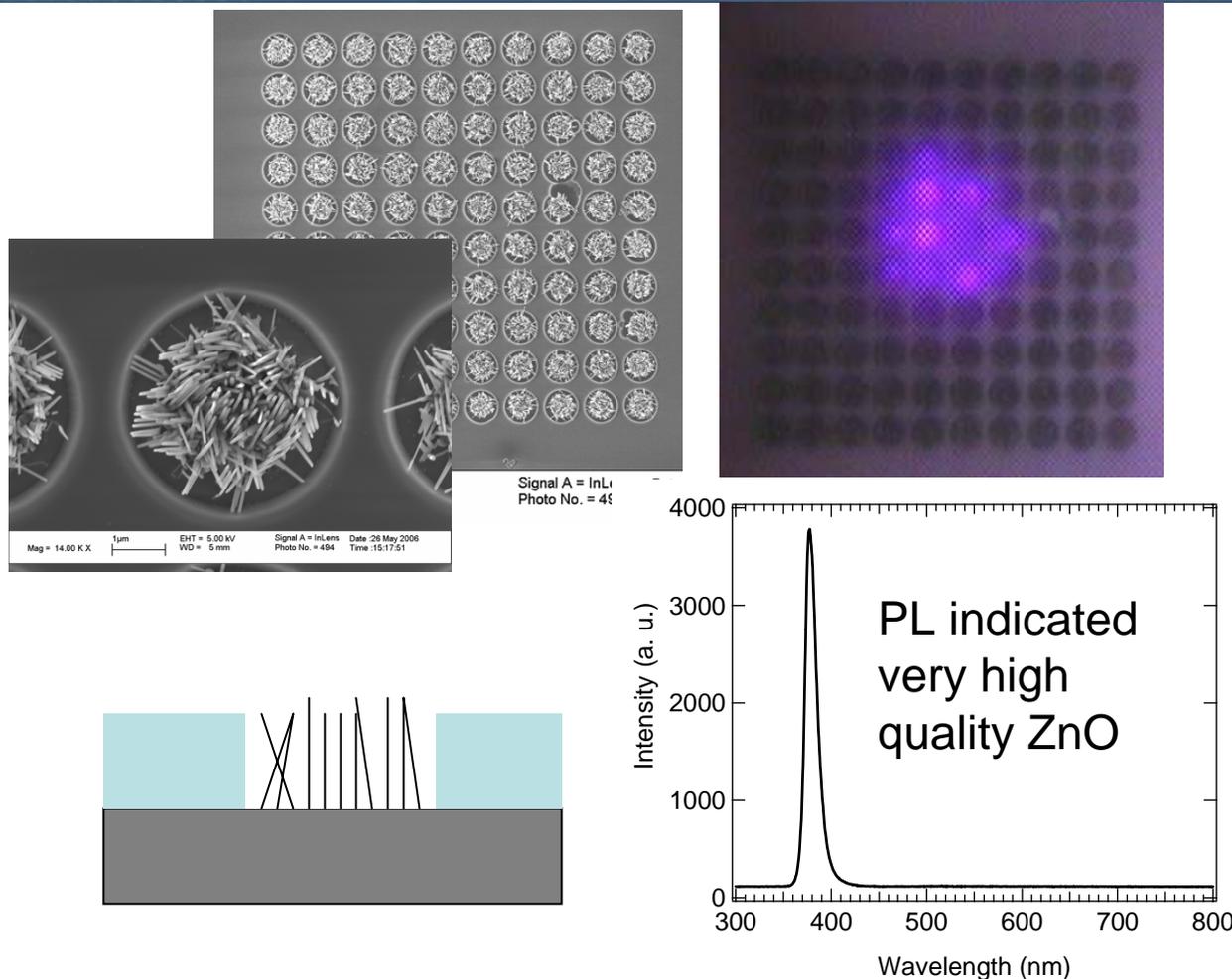
contacts to CNTs, nanowires



Leonard and Talin, PRL 97, 2006



Vertical integration: patterned VLS growth of ZnO nanowires in vertical wells (SNL/UCB)

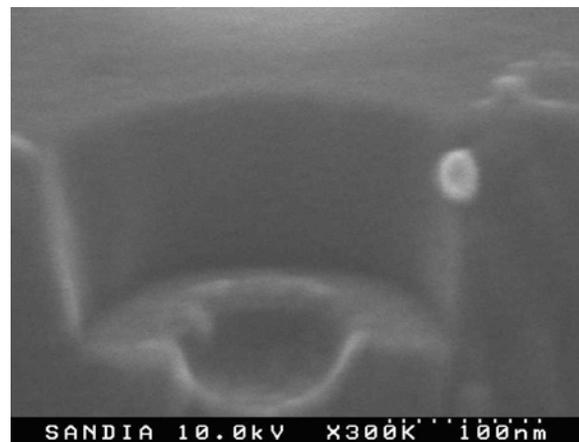
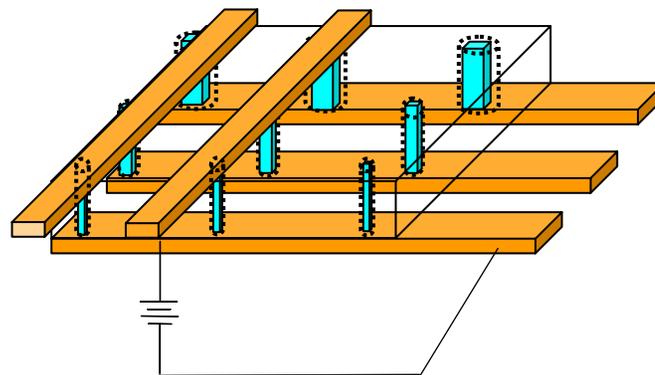
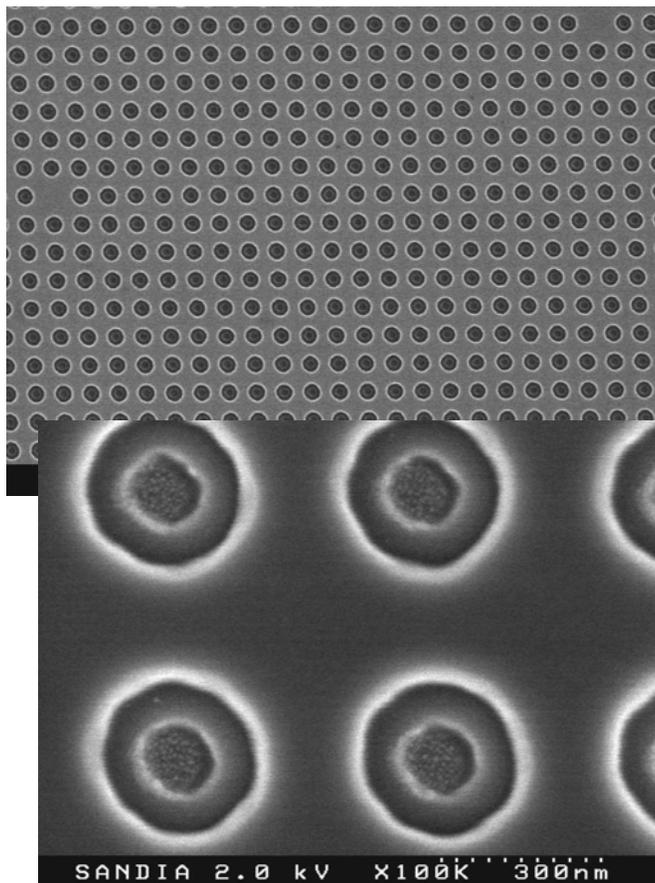


Patterned growth in vertical SiO₂ wells



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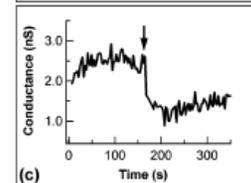
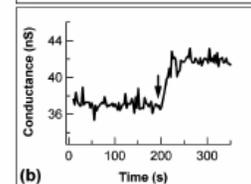
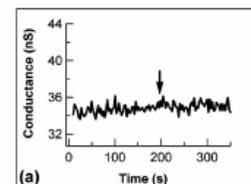
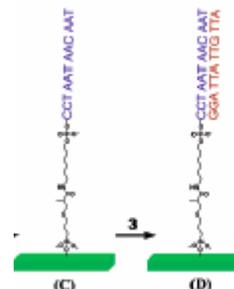
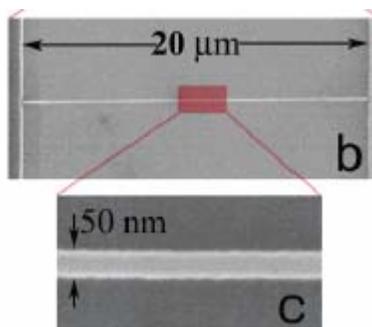
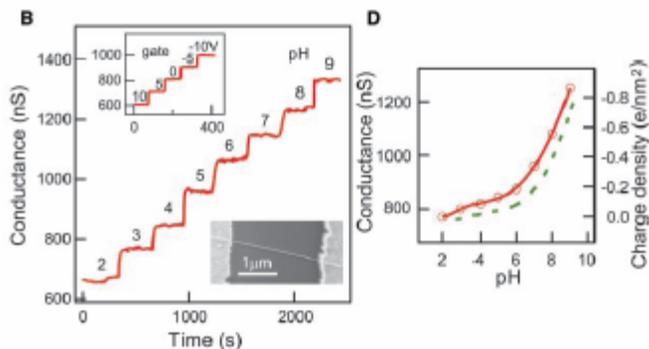
Vertical integration with dimensional, position control of catalyst



Nanowires for chemical and bio sensors

Nanowires exhibit excellent characteristics for sensors:

- Large surface/volume ratio.
- Electrical and optical properties are highly sensitive to the environment.
- Can be functionalized with chemically- and bio-active molecules.
- *Nanowires synthesized in bulk are difficult to integrate with input/output circuitry--need a practical, robust, reproducible method*



Real-time detection of DNA hybridization w/o labels using Si nanowire, *R. S. Williams group, Nanoletters 4, 245 (2004)*

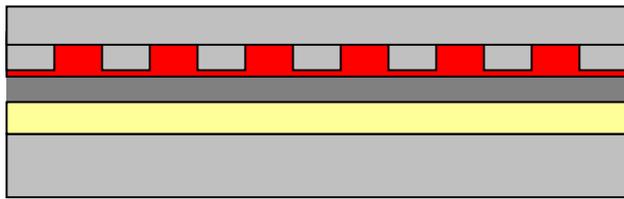
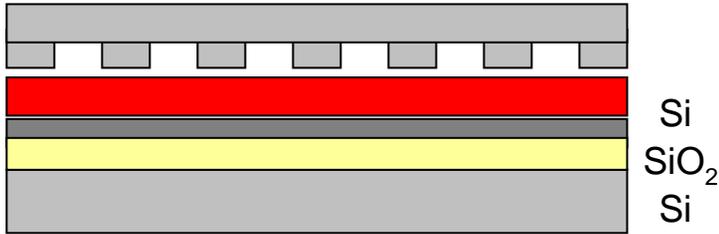
pH sensor fabricated by casting Si nanowires from sol'n, C. Lieber group, *Science 293 1289 (2001)*

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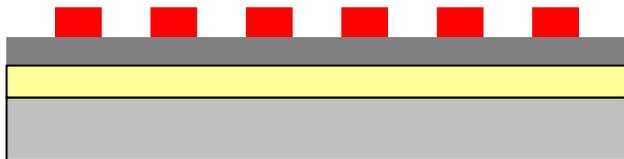


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Device Fabrication



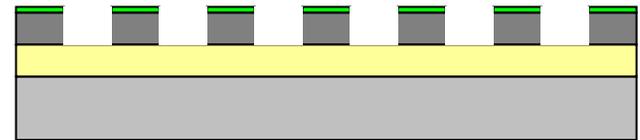
Thermally imprint



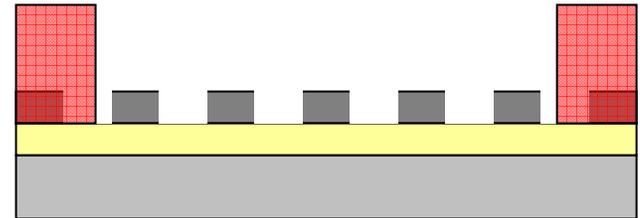
O₂ break through



Cr dep and lift-off



Cl₂ + HBr etch into Si device layer



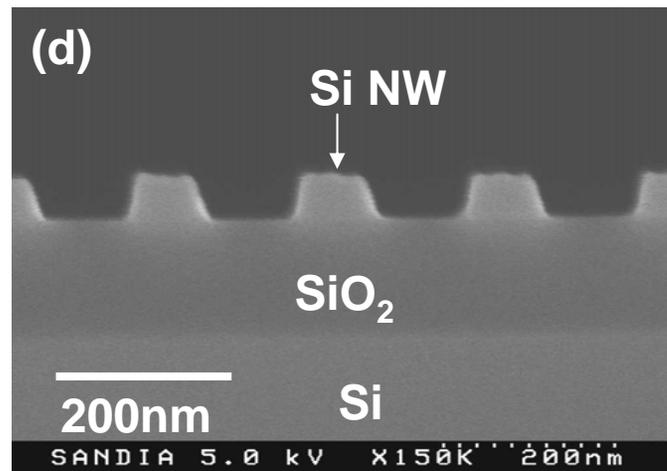
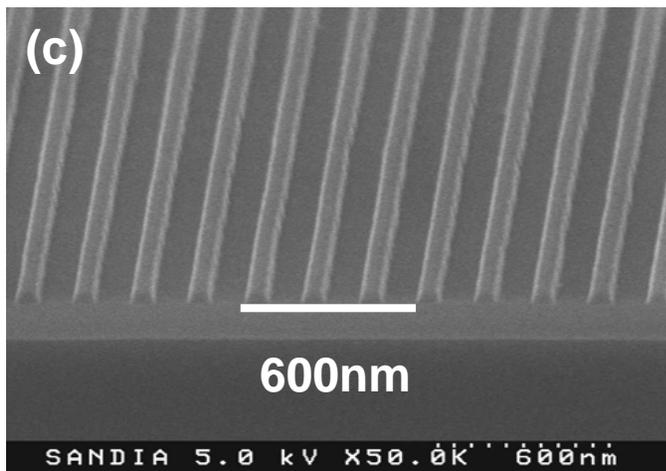
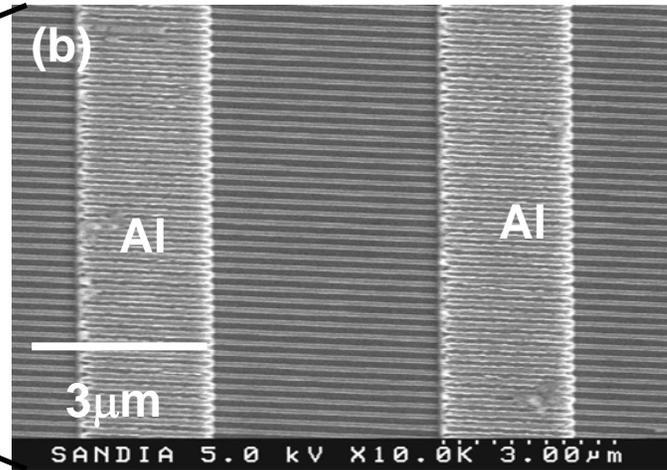
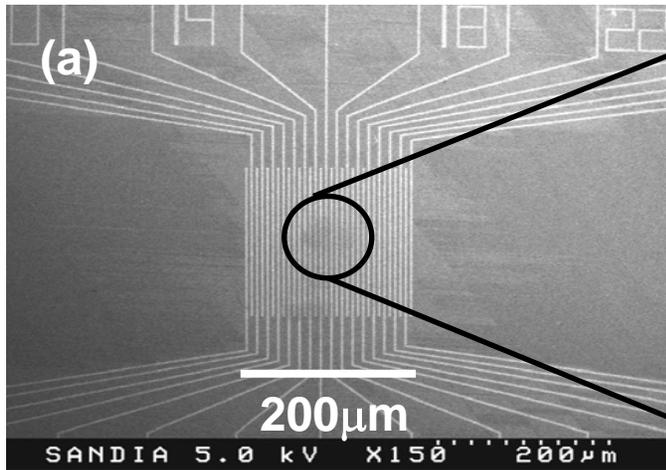
UV litho



Al dep and lift-off



Fabrication Results



Talin et al., Appl. Phys. Lett., in press



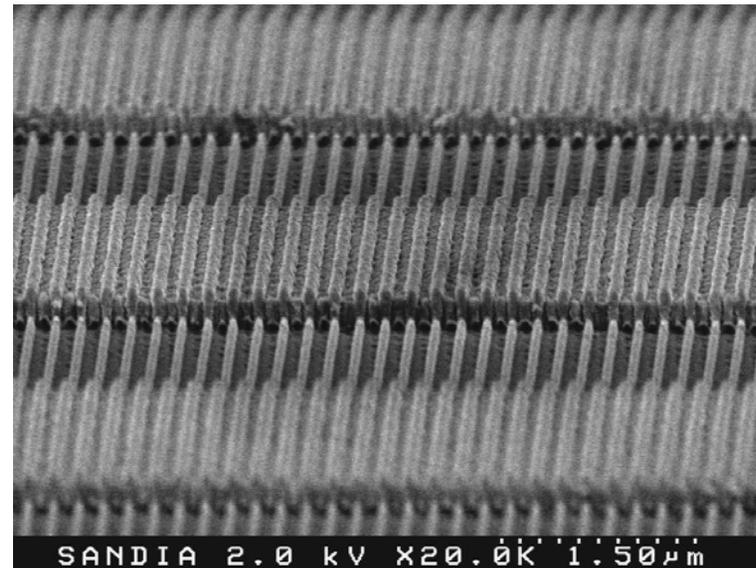
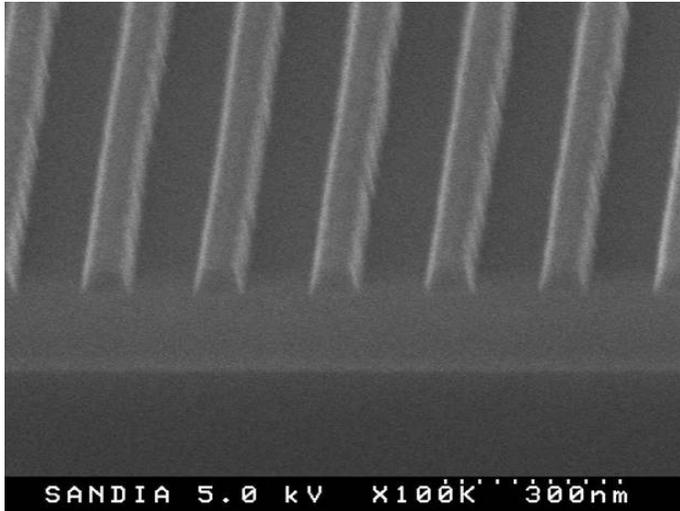
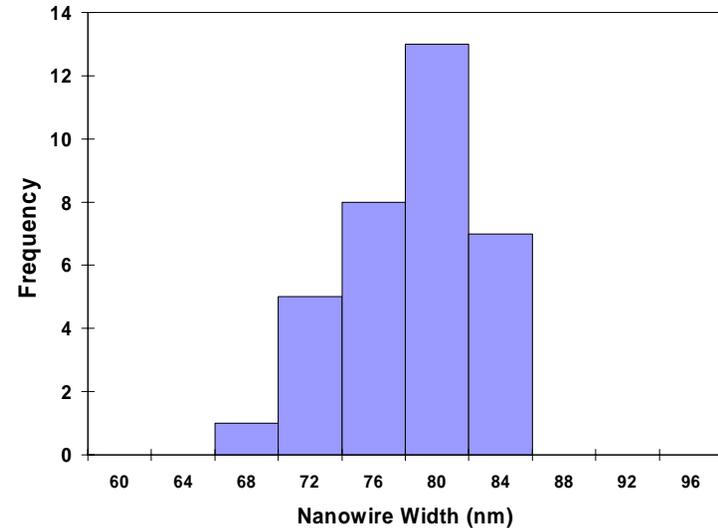
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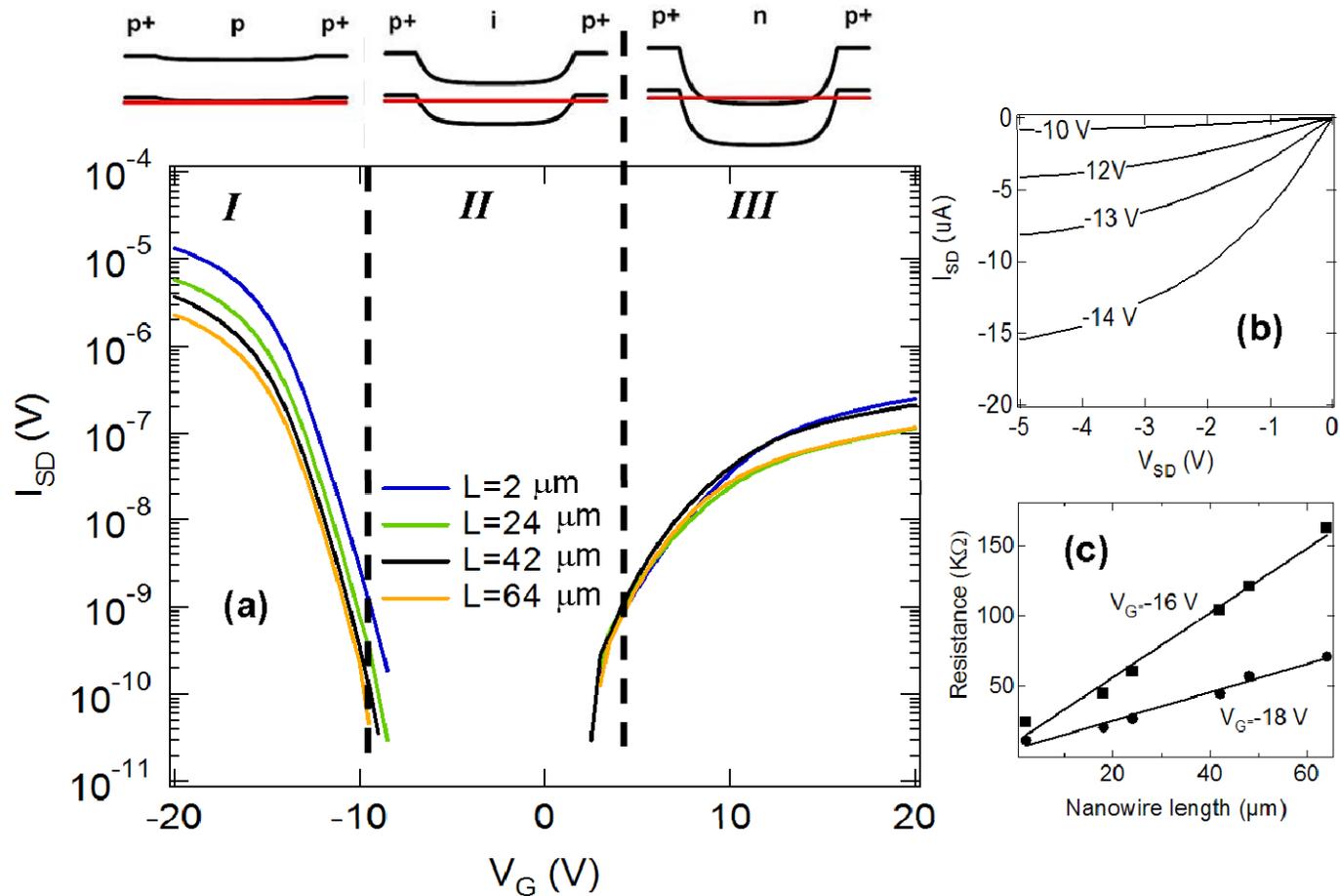
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Fabrication Results

- 400 devices on a 4" wafer
- 76 nm average line width
- Narrow distribution of widths



Electrical Characteristics

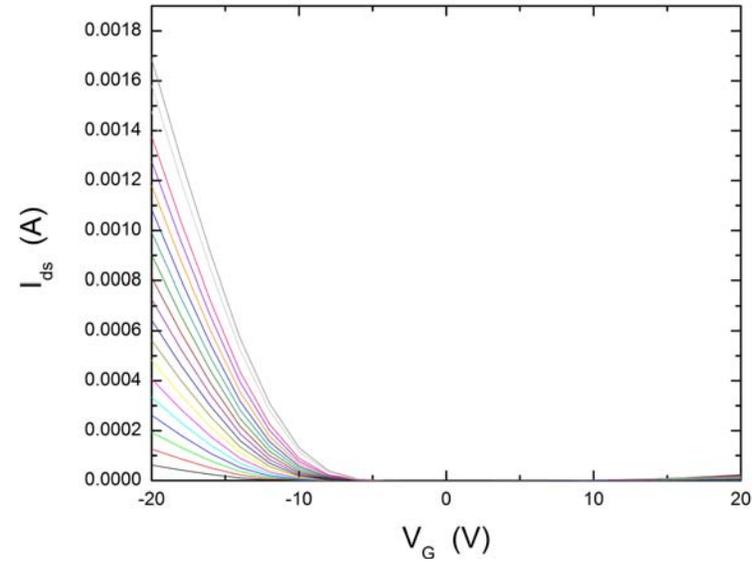


Talin et al., Appl. Phys. Lett., in press



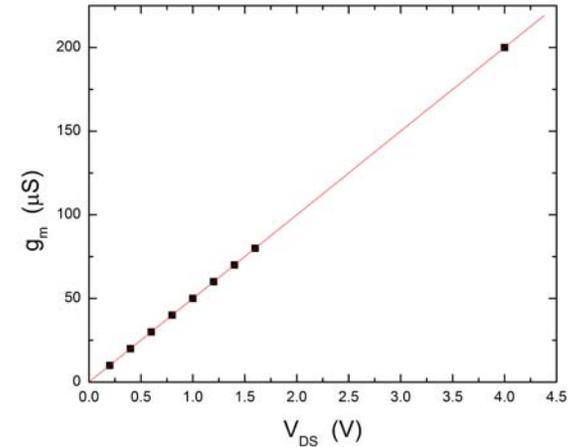
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Transconductance



$$g_m = \frac{\partial I_{ds}}{\partial V_g}$$

$$g_m = \mu C \frac{NW}{L} V_{DS}$$



$$C = 2\pi\epsilon L / \left[2\sum_{n=1}^N \ln\left(2\sqrt{h^2 + n^2l^2} / r\right) + \ln(2h/r) \right]$$

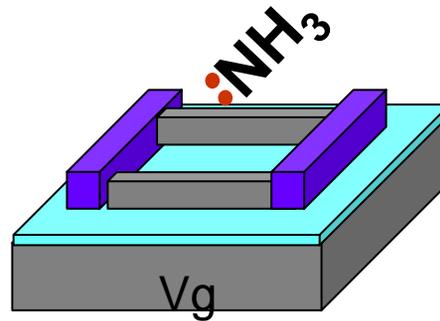
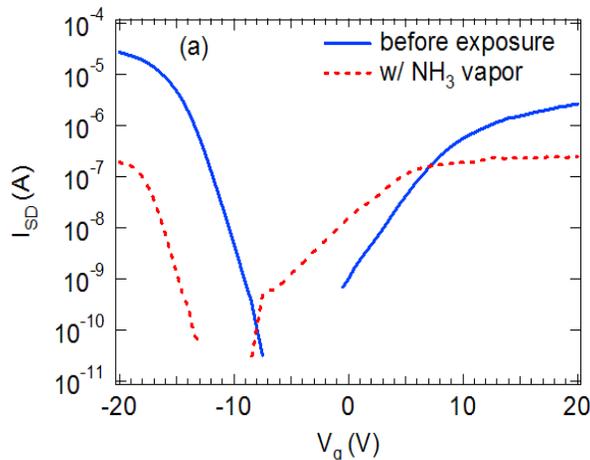
$$\mu = \frac{L}{\text{slope} * NWC} \approx 80 \text{ cm}^2/\text{Vs}$$

- Si hole mobility is $\sim 500 \text{ cm}^2/\text{Vs}$ (Sze)
- VLS Si nanowire mobility $0.006 - 3 \text{ cm}^2/\text{Vs}$ for 70 nm diam. (Lieber group, *J. Phys. Chem. B* 104, 5213, 2000)

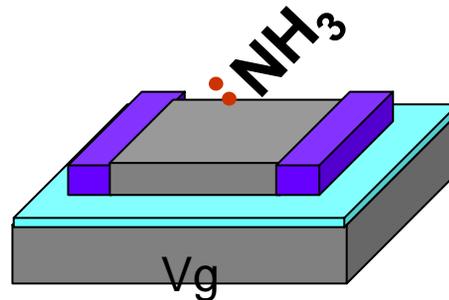
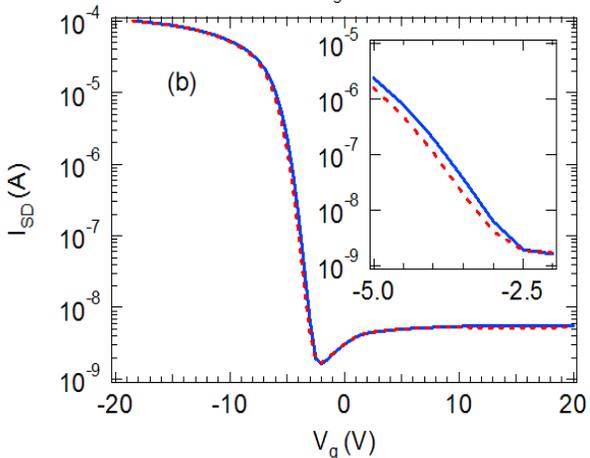


Sensors: nanowires more sensitive to ammonia vapor than thin films

Adsorption of electron donating ammonia makes Si nanowires more 'n-type', shifts $I_{SD}-V_g$ to the left



$$env = AC(V_t - V_g)$$

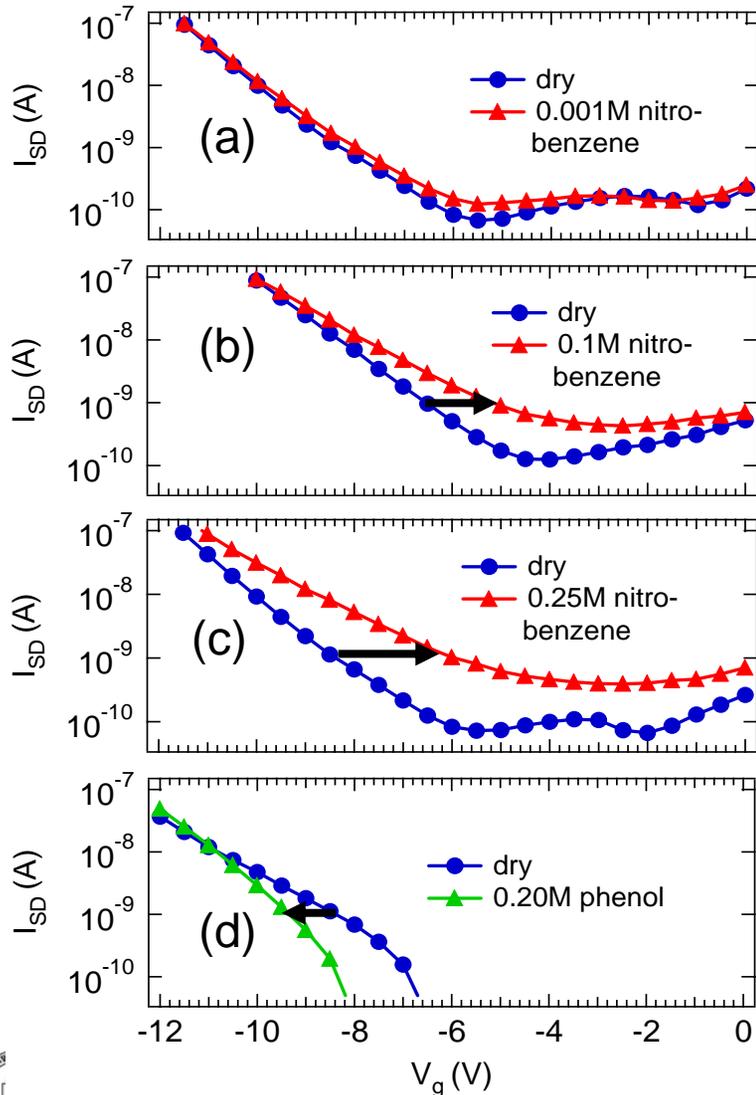


$$env + e\alpha\alpha^{-1}\theta s = AC(V_t' - V_g)$$

$$\Delta V_t = e\alpha\alpha^{-1}\theta C^{-1}(s/A)$$

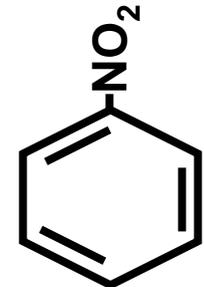
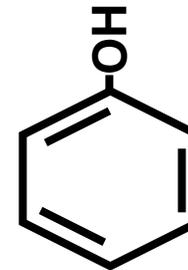
$$C \propto (l + \kappa w)/(l + w)$$

Sensing with Si nanowires: nitrobenzene, phenol



e^- donating

e^- withdrawing

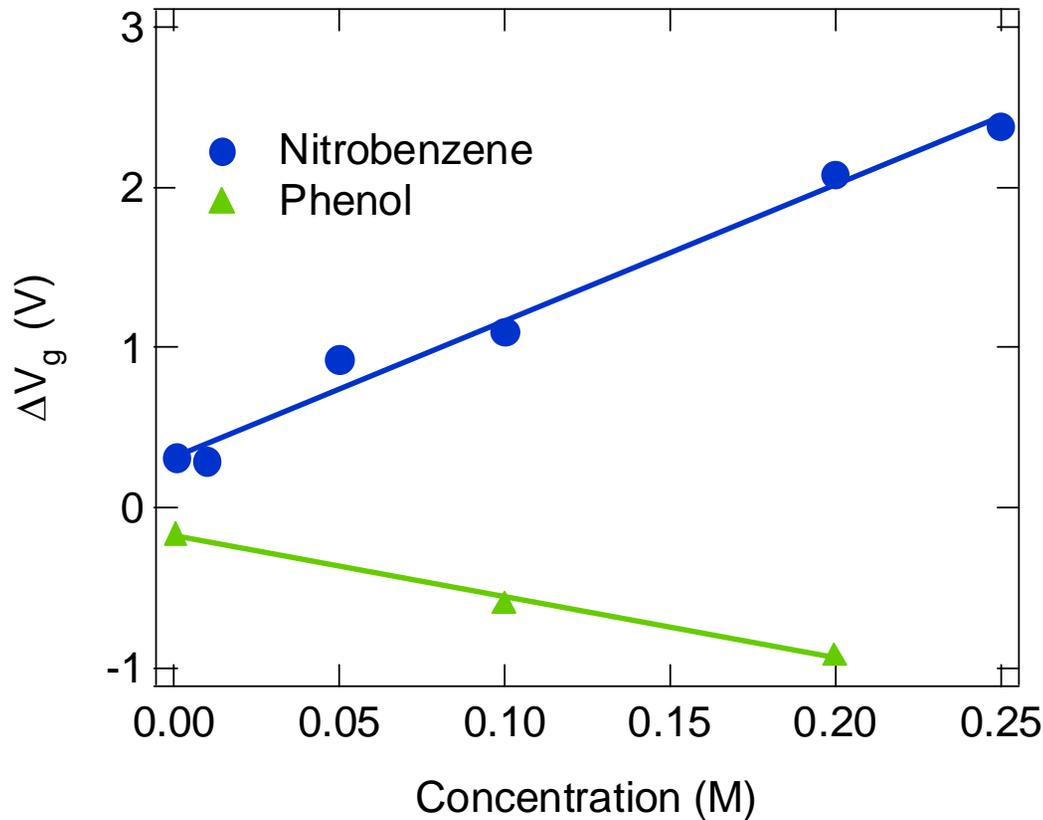


Hammett param. -0.4

$+0.78$

Sensing with Si nanowires: nitrobenzene, phenol

- *shift in V_g is linear with solute concentration*
- *scales with Hammett parameter (2:1 ratio for NB:Phenol)*



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